Food and Drug Administration, HHS

(c) To assure safe use of the additive, the label of the food additive container shall bear, in addition to other information required by the act, the name of the additive as follows:

(1) The additive identified in paragraph (a)(1) of this section as “partially defatted, cooked cottonseed flour”.

(2) The additive identified in paragraph (a)(2) of this section as “defatted cottonseed flour”.

(3) The additive identified in paragraph (a)(3) of this section as “roasted glandless cottonseed kernels”.

(4) The additive identified in paragraph (a)(4) of this section as “raw glandless cottonseed kernels for use in cooked hard candy”.

(d) The Food and Drug Administration and the Environmental Protection Agency have determined that glandless cottonseed kernels permitted for use by this section are a distinct commodity from glanded cottonseed.

§ 172.896 Dried yeasts.

Dried yeast (Saccharomyces cerevisiae and Saccharomyces fragilis) and dried torula yeast (Candida utilis) may be safely used in food provided the total folic acid content of the yeast does not exceed 0.04 milligram per gram of yeast (approximately 0.008 milligram of pteroyglutamic acid per gram of yeast).

§ 172.898 Bakers yeast glycan.

Bakers yeast glycan may be safely used in food in accordance with the following conditions:

(a) Bakers yeast glycan is the comminuted, washed, pasteurized, and dried cell walls of the yeast, Saccharomyces cerevisiae. It is composed principally of long chain carbohydrates, not less than 85 percent on a dry solids basis. The carbohydrate is composed of glucan and mannan units in approximately a 2:1 ratio.

(b) The additive meets the following specifications on a dry weight basis: Less than 0.4 part per million (ppm) arsenic, 0.13 ppm cadmium, 0.2 ppm lead, 0.05 ppm mercury, 0.09 ppm selenium, and 10 ppm zinc.

(c) The viable microbial content of the finished ingredient is:

<table>
<thead>
<tr>
<th>Use</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>(1) In salad dressings as an emulsifier and emulsifier salt as defined in §170.3(o)(8) of this chapter, stabilizer and thickener as defined in §170.3(o)(28) of this chapter, or texturizer as defined in §170.3(o)(32) of this chapter.</td>
<td>Not to exceed a concentration of 5 percent of the finished salad dressing.</td>
</tr>
<tr>
<td>(2) In frozen dessert analogs as a stabilizer and thickener as defined in §170.3(o)(28) of this chapter, or texturizer as defined in §170.3(o)(32) of this chapter.</td>
<td>In an amount not to exceed good manufacturing practice.</td>
</tr>
<tr>
<td>(3) In sour cream analogs as a stabilizer and thickener as defined in §170.3(o)(28) of this chapter, or texturizer as defined in §170.3(o)(32) of this chapter.</td>
<td>Do.</td>
</tr>
<tr>
<td>(4) In cheese spread analogs as a stabilizer and thickener as defined in §170.3(o)(28) of this chapter, or texturizer as defined in §170.3(o)(32) of this chapter.</td>
<td>Do.</td>
</tr>
<tr>
<td>(5) In cheese-flavored and sour cream-flavored snack dips as a stabilizer and thickener as defined in §170.3(o)(28) of this chapter, or texturizer as defined in §170.3(o)(32) of this chapter.</td>
<td>Do.</td>
</tr>
</tbody>
</table>

(e) The label and labeling of the ingredient shall bear adequate directions to assure that use of the ingredient complies with this regulation.


PART 173—SECONDARY DIRECT FOOD ADDITIVES PERMITTED IN FOOD FOR HUMAN CONSUMPTION

Subpart A—Polymer Substances and Polymer Adjuvants for Food Treatment

Sec.
173.5 Acrylate-acrylamide resins.
173.10 Modified polyacrylamide resin.
§ 173.5 Acrylate-acrylamide resins.

Acrylate-acrylamide resins may be safely used in food under the following prescribed conditions:

(a) The additive consists of one of the following:

(1) Acrylamide-acrylic acid resin (hydrolyzed polyacrylamide) is produced by the polymerization of acrylamide with partial hydrolysis, or by copolymerization of acrylamide and acrylic acid, with the greater part of the polymer being composed of acrylamide units.

(2) Sodium polyacrylate-acrylamide resin is produced by the polymerization and subsequent hydrolysis of acrylonitrile in a sodium silicate-sodium hydroxide aqueous solution, with the greater part of the polymer being composed of acrylamide units.

(b) The additive contains not more than 0.05 percent of residual monomer calculated as acrylamide.

(c) The additive is used or intended for use as follows:

(1) The additive identified in paragraph (a) (1) of this section is used as a flocculent in the clarification of beet sugar juice and liquor or cane sugar juice and liquor or corn starch hydrolyzate in an amount not to exceed 5 parts per million by weight of the juice or 10 parts per million by weight of the liquor or the corn starch hydrolyzate.
(2) The additive identified in paragraph (a)(2) of this section is used to control organic and mineral scale in beet sugar juice and liquor or cane sugar juice and liquor in an amount not to exceed 2.5 parts per million by weight of the juice or liquor.


§ 173.10 Modified polyacrylamide resin.

Modified polyacrylamide resin may be safely used in food in accordance with the following prescribed conditions:

(a) The modified polyacrylamide resin is produced by the copolymerization of acrylamide with not more than 5 mole percent β-methacryloxyethyltrimethylammonium methyl sulfate.

(b) The modified polyacrylamide resin contains not more than 0.05 percent residual acrylamide.

(c) The modified polyacrylamide resin is used as a flocculent in the clarification of beet or cane sugar juice in an amount not exceeding 5 parts per million by weight of the juice.

(d) To assure safe use of the additive, the label and labeling of the additive shall bear, in addition to the other information required by the act, adequate directions to assure use in compliance with paragraph (c) of this section.

§ 173.20 Ion-exchange membranes.

Ion-exchange membranes may be safely used in the processing of food under the following prescribed conditions:

(a) The ion-exchange membrane is prepared by subjecting a polystyrene base conforming to §177.1520 of this chapter to polymerization with styrene until the polystyrene phase of the base is not less than 16 percent nor more than 30 percent by weight. The base is then modified by reaction with chloromethyl methyl ether, and by subsequent amination with trimethylamine, dimethylamine, diethyleneetriamine, or dimethylethanolamine.

(b) The ion-exchange membrane is manufactured so as to comply with the following extraction limitations when subjected to the described procedure: Separate square-foot samples of membrane weighing approximately 14 grams each are cut into small pieces and refluxed for 4 hours in 150 cubic centimeters of the following solvents: Distilled water, 5 percent acetic acid, and 50 percent alcohol. Extraction from each sample will not exceed 0.4 percent by weight of sample.

(c) The ion-exchange membrane will be used in the production of grapefruit juice to adjust the ratio of citric acid to total solids of the grapefruit juice produced.

§ 173.21 Perfluorinated ion exchange membranes.

Substances identified in paragraph (a) of this section may be safely used as ion exchange membranes intended for use in the treatment of bulk quantities of liquid food under the following prescribed conditions:

(a) Identity. The membrane is a copolymer of ethanesulfonyl fluoride, 2-[1-difluoro-[(trifluoroethenyl)oxy]methyl]-1,2,2,2-tetrafluoroethoxy]-1,1,2,2-tetrafluoro-, with tetrafluoroethylene that has been subsequently treated to hydrolyze the sulfonyl fluoride group to the sulfonic acid. The Chemical Abstracts Service name of this polymer is ethanesulfonic acid, 2-[1-difluoro-[(trifluoroethenyl)oxy]methyl]-1,2,2,2-tetrafluoroethoxy]-1,1,2,2-tetrafluoro-, polymer with tetrafluoroethane (CAS Reg. No. 31175-20-9).

(b) Optional adjuvant substances. The basic polymer identified in paragraph (a) of this section may contain optional adjuvant substances required in the production of such basic polymer. These optional adjuvant substances may include substances used in accordance with §174.5 of this chapter.

(c) Conditions of use. (1) Perfluorinated ion exchange membranes described in paragraph (a) of this section may be used in contact with all types of liquid foods at temperatures not exceeding 70 °F (158 °F).

(2) Maximum thickness of the copolymer membrane is 0.007 inch (0.017 centimeter).

(3) Perfluorinated ion exchange membranes shall be maintained in a sanitary manner in accordance with current good manufacturing practice so as
§ 173.25 Ion-exchange resins.

Ion-exchange resins may be safely used in the treatment of food under the following prescribed conditions:

(a) The ion-exchange resins are prepared in appropriate physical form, and consist of one or more of the following:

(1) Sulfonated copolymer of styrene and divinylbenzene.

(2) Sulfonated anthracite coal meeting the requirements of ASTM method D 388-38, Class 1, Group 2, "Standard Specifications for Classification of Coal by Rank," which is incorporated by reference. Copies are available from University Microfilms International, 300 N. Zeeb Rd., Ann Arbor, MI 48106, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

(3) Sulfite-modified cross-linked phenol-formaldehyde, with modification resulting in sulfonic acid groups on side chains.

(4) Methacrylic acid-divinylbenzene copolymer.

(5) Cross-linked polystyrene, first chloromethylated then aminated with trimethylamine, dimethylamine, diethyleneetriamine, or dimethylethanolamine.

(6) Diethyleneetriamine, triethylene tetramine, or tetraethylenepentamine cross-linked with epichlorohydrin.

(7) Cross-linked phenol-formaldehyde activated with one or both of the following: Triethylene tetramine and tetraethylenepentamine.

(8) Reaction resin of formaldehyde, acetone, and tetraethylenepentamine.

(9) Completely hydrolyzed copolymers of methyl acrylate and divinylbenzene.

(10) Completely hydrolyzed terpolymers of methyl acrylate, divinylbenzene, and acrylonitrile.

(11) Sulfonated terpolymers of styrene, divinylbenzene, and acrylonitrile or methyl acrylate.

(12) Methyl acrylate-divinylbenzene copolymer containing not less than 2 percent by weight of divinylbenzene, aminolyzed with dimethylaminopropylamine.

(13) Methyl acrylate-divinylbenzene copolymer containing not less than 3.5 percent by weight of divinylbenzene, aminolyzed with dimethylaminopropylamine.

(14) Epichlorohydrin cross-linked with ammonia.

(15) Sulfonated tetrapolymer of styrene, divinylbenzene, acrylonitrile, and methyl acrylate derived from a mixture of monomers containing not more than a total of 2 percent by weight of acrylonitrile and methyl acrylate.

(16) Methyl acrylate-divinylbenzene-diethylene glycol divinyl ether terpolymer containing not less than 3.5 percent by weight of divinylbenzene and not more than 0.6 percent by weight of diethylene glycol divinyl ether, aminolyzed with dimethylaminopropylamine.

(17) Styrene-divinylbenzene cross-linked copolymer, first chloromethylated then aminated with dimethylamine and oxidized with hydrogen peroxide whereby the resin contains not more than 15 percent by weight of vinyl N,N-dimethylbenzylamine-N-oxide and not more than 6.5 percent by weight of nitrogen.

(18) Methyl acrylate-divinylbenzene-diethylene glycol divinyl ether terpolymer containing not less than 7 percent by weight of divinylbenzene and not more than 2.3 percent by weight of diethylene glycol divinyl ether, aminolyzed with dimethylaminopropylamine and quaternized with methyl chloride.

(19) Epichlorohydrin cross-linked with ammonia and then quaternized with methyl chloride to contain not more than 18 percent strong base capacity by weight of total exchange capacity (Chemical Abstracts Service name: Oxirane (chloromethyl)-, polymer with ammonia, reaction product with chloromethane; CAS Reg. No. 68036-99-7).
(20) Regenerated cellulose, cross-linked and alkylated with epichlorohydrin and propylene oxide, then sulfonated whereby the amount of epichlorohydrin plus propylene oxide employed does not exceed 250 percent by weight of the starting quantity of cellulose.

(b) Ion-exchange resins are used in the purification of foods, including potable water, to remove undesirable ions or to replace less desirable ions with one or more of the following: bicarbonate, calcium, carbonate, chloride, hydrogen, hydroxyl, magnesium, potassium, sodium, and sulfate except that: The ion-exchange resin identified in paragraph (a)(12) of this section is used only in accordance with paragraph (b)(1) of this section, the ion-exchange resin identified in paragraph (a)(13) of this section is used only in accordance with paragraph (b)(2) of this section, the resin identified in paragraph (a)(16) of this section is used only in accordance with paragraph (b)(3) of this section, and the ion-exchange resin identified in paragraph (a)(17) of this section is used only in accordance with paragraph (b)(4) of this section, but the strongly acidic cation-exchange resins referred to in paragraphs (b)(1) (i) and (ii) of this section used in the process meet the requirements of paragraph (c)(4) of this section, except for the exemption described in paragraph (d) of this section.

(2) The ion-exchange resins identified in paragraphs (a)(13) and (16) of this section are used to treat water and aqueous food only of the types identified under Categories I, II, and VI-B in table 1 of §176.170(c) of this chapter: Provided, That the temperature of the water or food passing through the resin beds is maintained at 50 °C or less and the flow rate of the water or food passing through the beds is not less than 0.5 gallon per cubic foot per minute.

(i) The ion-exchange resin identified in paragraph (a)(13) of this section is used to treat water and aqueous food only of the types identified under Categories I, II, and VI-B in Table 1 of §176.170(c) of this chapter, Provided, That the temperature of the water or food passing through the resin bed is maintained at 50 °C or less and the flow rate of the water or food passing through the bed is not less than 0.5 gallon per cubic foot per minute.

(ii) The ion-exchange resin identified in paragraph (a)(16) of this section is used to treat water and aqueous food only of the types identified under Categories I, II, and VI-B in Table 1 of §176.170(c) of this chapter, Provided, that either:

(A) The temperature of the water or food passing through the resin bed is maintained at 50 °C or less and the flow rate of the water or food passing through the bed is not less than 0.5 gallon per cubic foot per minute; or
(B) Extracts of the resin will be found to contain no more than 1 milligram/kilogram dimethylaminopropylamine in each of the food simulants, distilled water and 10 percent ethanol, when, following washing and pretreatment of the resin in accordance with §173.25(c)(1), the resin is subjected to the following test under conditions simulating the actual temperature and flow rate of use: “The Determination of 3-Dimethylaminopropylamine in Food Simulating Extracts of Ion Exchange Resins,” February 4, 1998, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the Division of Petition Control (HFS-215), Center for Food Safety and Applied Nutrition, Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or may be examined at the Center for Food Safety and Applied Nutrition’s Library, 200 C St. SW., rm. 3321, Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC.

(3) The ion-exchange resin identified in paragraph (a)(17) of this section is used only for industrial application to treat bulk quantities of aqueous food, including potable water, or for treatment of municipal water supplies, subject to the condition that the temperature of the food or water passing through the resin bed is maintained at 25 °C or less and the flow rate of the food or water passing through the bed is not less than 2 gallons per cubic foot per minute.

(4) The ion-exchange resin identified in paragraph (a)(18) of this section is used to treat aqueous sugar solutions subject to the condition that the temperature of the sugar solution passing through the resin bed is maintained at 82 °C (179.6 °F) or less and the flow rate of the sugar solution passing through the bed is not less than 46.8 liters per cubic meter (0.35 gallon per cubic foot) of resin bed volume per minute.

(5) The ion-exchange resin identified in paragraph (a)(20) of this section is limited to use in aqueous process streams for the isolation and purification of protein concentrates and isolates under the following conditions:

(i) For resins that comply with the requirements in paragraph (d)(2)(i) of this section, the pH range for the resin shall be no less than 3.5 and no more than 9, and the temperatures of water and food passing through the resin bed shall not exceed 25°C.

(ii) For resins that comply with the requirements in paragraph (d)(2)(ii) of this section, the pH range for the resin shall be no less than 2 and no more than 10, and the temperatures of water and food passing through the resin shall not exceed 50 °C.

(c) To insure safe use of ion-exchange resins, each ion-exchange resin will be:

(1) Subjected to pre-use treatment by the manufacturer and/or the user in accordance with the manufacturer’s directions prescribed on the label or labeling accompanying the resins, to guarantee a food-grade purity of ion-exchange resins, in accordance with good manufacturing practice.

(2) Accompanied by label or labeling to include directions for use consistent with the intended functional purpose of the resin.

(3) Used in compliance with the label or labeling required by paragraph (c)(2) of this section.

(4) Found to result in no more than 1 part per million of organic extractives obtained with each of the named solvents, distilled water, 15 percent alcohol, and 5 percent acetic acid when, having been washed and otherwise treated in accordance with the manufacturer’s directions for preparing them for use with food, the ion-exchange resin is subjected to the following test: Using a separate ion-exchange column for each solvent, prepare columns using 50 milliliters of the ready to use ion-exchange resin that is to be tested. While maintaining the highest temperature that will be encountered in use pass through these beds at the rate of 350–450 milliliters per hour the three test solvents distilled water, 15 percent (by volume) ethyl alcohol, and 5 percent (by weight) acetic acid. The first liter of effluent from each solvent is discarded, then the next 2 liters are used to determine organic extractives. The 2-liter sample is carefully evaporated to constant weight at 105 °C; this is total extractives. This residue is fired in a muffle furnace at 850 °C to constant weight; this is ash. Total extractives, minus...
ash equals the organic extractives. If the organic extractives are greater than 1 part per million of the solvent used, a blank should be run on the solvent and a correction should be made by subtracting the total extractives obtained with the blank from the total extractives obtained in the resin test. The solvents used are to be made as follows:

Distilled water (de-ionized water is distilled). 15 percent ethyl alcohol made by mixing 15 volumes of absolute ethyl alcohol A.C.S. reagent grade, with 85 volumes of distilled de-ionized water.

5 percent acetic acid made by mixing 5 parts by weight of A.C.S. reagent grade glacial acetic acid with 95 parts by weight of distilled de-ionized water.

In addition to the organic extractives limitation prescribed in this paragraph, the ion-exchange resin identified in paragraph (a)(17) of this section, when extracted with each of the named solvents, distilled water, 50 percent alcohol, and 5 percent acetic acid, will be found to result in not more than 7 parts per million of nitrogen extractives (calculated as nitrogen) when the resin in the free-base form is subjected to the following test immediately before each use: Using a separate 1-inch diameter glass ion-exchange column for each solvent, prepare each column using 100 milliliters of ready to use ion-exchange resin that is to be tested. With the bottom outlet closed, fill each ion-exchange column with one of the three solvents at a temperature of 25°C until the solvent level is even with the top of the resin bed. Seal each column at the top and bottom and store in a vertical position at a temperature of 25°C. After 96 hours, open the top of each column, drain the solvent into a collection vessel, and analyze each drained solvent and a solvent blank for nitrogen by a standard micro-Kjeldahl method.

(d)(1) The ion-exchange resins identified in paragraphs (a)(1), (a)(2), (a)(11), and (a)(15) of this section are exempted from the acetic acid extraction requirement of paragraph (c)(4) of this section.

(ii) The extraction requirement in paragraph (c)(4) of this section by using reagent grade hydrochloric acid, diluted to pH 2, as a substitute for acetic acid. The resin shall be found to result in no more than 25 parts per million of organic extractives obtained with each of the following solvents: Distilled water; 15 percent alcohol; and hydrochloric acid, pH 2. Blanks should be run for each of the solvents, and corrections should be made by subtracting the total extractives obtained with the blank from the total extractives obtained in the resin test.

(e) Acrylonitrile copolymers identified in this section shall comply with the provisions of §180.22 of this chapter.

§ 173.40 Molecular sieve resins.

Molecular sieve resins may be safely used in the processing of food under the following prescribed conditions:

(a) The molecular sieve resins consist of purified dextran having an average molecular weight of 40,000, cross-linked with epichlorohydrin in a ratio of 1 part of dextran to 10 parts of epichlorohydrin, to give a stable three dimensional structure. The resins have a pore size of 2.0 to 3.0 milliliters per gram of dry resin (expressed in terms of water regain), and a particle size of 10 to 300 microns.

(b) The molecular sieve resins are thoroughly washed with potable water prior to their first use in contact with food.

(c) Molecular sieve resins are used as the gel filtration media in the final purification of partially delactosed whey. The gel bed shall be maintained in a sanitary manner in accordance with good manufacturing practice so as to prevent microbial build-up on the bed and adulteration of the product.
§ 173.45 Polymaleic acid and its sodium salt.

Polymaleic acid (CAS Reg. No. 26099-09-2) and its sodium salt (CAS Reg. No. 70247-90-4) may be safely used in food in accordance with the following prescribed conditions:

(a) The additives have a weight-average molecular weight in the range of 540 to 850 and a number-average molecular weight in the range of 520 to 650, calculated as the acid. Molecular weights shall be determined by a method entitled “Determination of Molecular Weight Distribution of Poly(Maleic) Acid,” March 17, 1992, produced by Ciba-Geigy, Inc., Seven Skyline Dr., Hawthorne, NY 10532-2188, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the Division of Product Policy, Center for Food Safety and Applied Nutrition (HFS-205), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or are available for inspection at the Center for Food Safety and Applied Nutrition’s Library, 200 C St. SW., rm. 3321, Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC.

(b) The additives may be used, individually or together, in the processing of beet sugar juice and liquor or of cane sugar juice and liquor to control mineral scale.

(c) The additives are to be used so that the amount of either or both additives does not exceed 4 parts per million (calculated as the acid) by weight of the beet or cane sugar juice or liquor process stream.


§ 173.50 Polyvinylpolypyrrolidone.

The food additive polyvinylpolypyrrolidone may be safely used in accordance with the following prescribed conditions:

(a) The additive is a homopolymer of purified vinylpyrrolidone catalytically produced under conditions producing polymerization and cross-linking such that an insoluble polymer is produced.

(b) The food additive is so processed that when the finished polymer is refluxed for 3 hours with water, 5 percent acetic acid, and 50 percent alcohol, no more than 50 parts per million of extractables is obtained with each solvent.

(c) It is used or intended for use as a clarifying agent in beverages and vinegar, followed by removal with filtration.

§ 173.55 Polyvinylpyrrolidone.

The food additive polyvinylpyrrolidone may be safely used in accordance with the following prescribed conditions:

(a) The additive is a polymer of purified vinylpyrrolidone catalytically produced, having an average molecular weight of 40,000 and a maximum unsaturation of 1 percent, calculated as the monomer, except that the polyvinylpyrrolidone used in beer is that having an average molecular weight of 360,000 and a maximum unsaturation of 1 percent, calculated as the monomer.

(b) The additive is used or intended for use in foods as follows:

<table>
<thead>
<tr>
<th>Food</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Beer</td>
<td>As a clarifying agent, at a residual level not to exceed 10 parts per million.</td>
</tr>
<tr>
<td>Flavor concentrates in tablet form</td>
<td>As a tableting adjunct, in an amount not to exceed good manufacturing practice.</td>
</tr>
<tr>
<td>Non-nutritive sweeteners in concentrated liquid form</td>
<td>As a stabilizer, bodying agent, and dispersant, in an amount not to exceed good manufacturing practice.</td>
</tr>
<tr>
<td>Non-nutritive sweeteners in tablet form</td>
<td>As a tableting adjunct, in an amount not to exceed good manufacturing practice.</td>
</tr>
<tr>
<td>Vitamin and mineral concentrates in liquid form</td>
<td>As a stabilizer, bodying agent, and dispersant, in an amount not to exceed good manufacturing practice.</td>
</tr>
<tr>
<td>Vitamin and mineral concentrates in tablet form</td>
<td>As a tableting adjunct, in an amount not to exceed good manufacturing practice.</td>
</tr>
<tr>
<td>Vinegar</td>
<td>As a clarifying agent, at a residual level not to exceed 10 parts per million.</td>
</tr>
<tr>
<td>Wine</td>
<td>As a clarifying agent, at a residual level not to exceed 10 parts per million.</td>
</tr>
</tbody>
</table>
§ 173.60 Dimethylamine-epichlorohydrin copolymer.

Dimethylamine-epichlorohydrin copolymer (CAS Reg. No. 25988–97–0) may be safely used in food in accordance with the following prescribed conditions:

(a) The food additive is produced by copolymerization of dimethylamine and epichlorohydrin in which not more than 5 mole-percent of dimethylamine may be replaced by an equimolar amount of ethylenediamine, and in which the mole ratio of total amine to epichlorohydrin is approximately 1:1.

(b) The additive meets the following specifications:

1. The nitrogen content of the copolymer is 9.4 to 10.8 weight percent on a dry basis.
2. A 50-percent-by-weight aqueous solution of the copolymer has a minimum viscosity of 175 centipoises at 25 °C as determined by LVT-series Brookfield viscometer using a No. 2 spindle at 60 RPM (or by another equivalent method).
3. The additive contains not more than 1,000 parts per million of 1,3-dichloro-2-propanol and not more than 10 parts per million epichlorohydrin. The epichlorohydrin and 1,3-dichloro-2-propanol content is determined by an analytical method entitled “The Determination of Epichlorohydrin and 1,3-Dichloro-2-Propanol in Dimethylamine-Epichlorohydrin Copolymer,” which is incorporated by reference. Copies are available from the Center for Food Safety and Applied Nutrition (HFS–200), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.
4. Heavy metals (as Pb), 2 parts per million maximum.
5. Arsenic (as As), 2 parts per million maximum.

(c) The food additive is used as a decolorizing agent and/or flocculant in the clarification of refinery sugar liquors and juices. It is added only at the defecation/clarification stage of sugar liquor refining at a concentration not to exceed 150 parts per million of copolymer by weight of sugar solids.

(d) To assure safe use of the additive, the label and labeling of the additive shall bear, in addition to other information required by the Act, adequate directions to assure use in compliance with paragraph (c) of this section.

§ 173.65 Divinylbenzene copolymer.

Divinylbenzene copolymer may be used for the removal of organic substances from aqueous foods under the following prescribed conditions:

(a) The copolymer is prepared in appropriate physical form and is derived by the polymerization of a grade of divinylbenzene which comprises at least 79 weight-percent divinylbenzene, 15 to 20 weight-percent ethylvinylbenzene, and no more than 4 weight-percent nonpolymerizable impurities.

(b) In accordance with the manufacturer’s directions, the copolymer described in paragraph (a) of this section is subjected to pre-use extraction with a water soluble alcohol until the level of divinylbenzene in the extract is less than 50 parts per billion as determined by a method titled, “The Determination of Divinylbenzene in Alcohol Extracts of Amberlite XAD-4,” which is incorporated by reference. Copies of this method are available from the Center for Food Safety and Applied Nutrition (HFS–200), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

The copolymer is then treated with water according to the manufacturer’s recommendation to remove the extraction solvent to guarantee a food-grade purity of the resin at the time of use, in accordance with current good manufacturing practice.

(c) The temperature of the aqueous food stream contacting the polymer is maintained at 79.4 °C (175 °F) or less.

(d) The copolymer may be used in contact with food only of Types I, II, and VI–B (excluding carbonated beverages) described in table 1 of paragraph (c) of §176.170 of this chapter.
§ 173.70 Chloromethylated aminated styrene-divinylbenzene resin.

Chloromethylated aminated styrene-divinylbenzene copolymer (CAS Reg. No. 60177-39-1) may be safely used in food in accordance with the following prescribed conditions:

(a) The additive is an aqueous dispersion of styrene-divinylbenzene copolymers, first chloromethylated then aminated with trimethylamine, having an average particle size of not more than 2.0 microns.

(b) The additive shall contain no more than 3.0 percent nonvolatile, soluble extractives when tested as follows: One hundred grams of the additive is centrifuged at 17,000 r/min for 2 hours. The resulting clear supernatant is removed from the compacted solids and concentrated to approximately 10 grams on a steam bath. The 10-gram sample is again centrifuged at 17,000 r/min for 2 hours to remove any residual insoluble material. The supernatant from the second centrifugation is then removed from any compacted solids and dried to constant residual weight using a steam bath. The percent nonvolatile solubles is obtained by dividing the weight of the dried residue by the weight of the solids in the original resin dispersion.

(c) The additive is used as a decolorizing and clarification agent for treatment of refinery sugar liquors and juices at levels not to exceed 500 parts of additive solids per million parts of sugar solids.

§ 173.70 Chloromethylated aminated styrene-divinylbenzene resin. (1) A weight average molecular weight of 2,000 to 2,300; and (2) A weight average molecular weight to number average molecular weight ratio of not more than 1.3. Copies of the method are available from the Center for Food Safety and Applied Nutrition (HFS-200), Food and Drug Administration, 200 C Street SW., Washington, DC 20204, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20402.

§ 173.75 Sorbitan monooleate.

Sorbitan monooleate may be safely used in food in accordance with the following prescribed conditions:

(a) The additive is produced by the esterification of sorbitol with commercial oleic acid.

(b) It meets the following specifications:

(1) Saponification number, 145-160.
(2) Hydroxyl number, 193-210.

(c) The additive is used or intended for use as follows:

(1) As an emulsifier in polymer dispersions that are used in the clarification of cane or beet sugar juice or liquor in an amount not to exceed 7.5 parts per million in sugar liquor.
(2) The additive is used in an amount not to exceed 0.70 part per million in sugar juice and 1.4 parts per million in sugar liquor.

§ 173.110 Amyloglucosidase derived from Rhizopus niveus.

Amyloglucosidase enzyme product, consisting of enzyme derived from Rhizopus niveus, and diatomaceous silica as a carrier, may be safely used in food in accordance with the following conditions:
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(a) Rhizopus niveus is classified as follows: Class, Phycomycetes; order, Mucorales; family, Mucoraceae; genus, Rhizopus; species, niveus.

(b) The strain of Rhizopus niveus is nonpathogenic and nontoxic in man or other animals.

(c) The enzyme is produced by a process which completely removes the organism Rhizopus niveus from the amyloglucosidase.

(d) The additive is used or intended for use for degrading gelatinized starch into constituent sugars, in the production of distilled spirits and vinegar.

(e) The additive is used at a level not to exceed 0.1 percent by weight of the gelatinized starch.

§ 173.120 Carbohydrase and cellulase derived from Aspergillus niger.

Carbohydrase and cellulase enzyme preparation derived from Aspergillus niger may be safely used in food in accordance with the following prescribed conditions:

(a) Aspergillus niger is classified as follows: Class, Deuteromycetes; order, Moniliales; family, Moniliaceae; genus, Aspergillus; species, niger.

(b) The strain of Aspergillus niger is nonpathogenic and nontoxic in man or other animals.

(c) The additive is produced by a process that completely removes the organism Aspergillus niger from the carbohydrate and cellulase enzyme product.

(d) The additive is used or intended for use as follows:

(1) For removal of visceral mass (belies) in clam processing.

(2) As an aid in the removal of the shell from the edible tissue in shrimp processing.

(e) The additive is used in an amount not in excess of the minimum required to produce its intended effect.

§ 173.130 Carbohydrase derived from Rhizopus oryzae.

Carbohydrase from Rhizopus oryzae may be safely used in the production of dextrose from starch in accordance with the following prescribed conditions:

(a) Rhizopus oryzae is classified as follows: Class, Phycomycetes; order, Mucorales; family, Mucoraceae; genus, Rhizopus; species, Rhizopus oryzae.

(b) The strain of Rhizopus oryzae is nonpathogenic and nontoxic.

(c) The carbohydrase is produced under controlled conditions to maintain nonpathogenicity and nontoxicity, including the absence of aflatoxin.

(d) The carbohydrase is produced by a process which completely removes the organism Rhizopus oryzae from the carbohydrase product.

(e) The carbohydrase is maintained under refrigeration from production to use and is labeled to include the necessity of refrigerated storage.

§ 173.135 Catalase derived from Micrococcus lysodeikticus.

Bacterial catalase derived from Micrococcus lysodeikticus by a pure culture fermentation process may be safely used in destroying and removing hydrogen peroxide used in the manufacture of cheese, in accordance with the following conditions.

(a) The organism Micrococcus lysodeikticus from which the bacterial catalase is to be derived is demonstrated to be nontoxic and nonpathogenic.

(b) The organism Micrococcus lysodeikticus is removed from the bacterial catalase prior to use of the bacterial catalase.

(c) The bacterial catalase is used in an amount not in excess of the minimum required to produce its intended effect.

§ 173.140 Esterase-lipase derived from Mucor miehei.

Esterase-lipase enzyme, consisting of enzyme derived from Mucor miehei var. Cooney et Emerson by a pure culture fermentation process, with maltodextrin or sweet whey as a carrier, may be safely used in food in accordance with the following conditions:

(a) Mucor miehei var. Cooney et Emerson is classified as follows: Class, Phycomycetes; subclass, Zygomycetes; order, Mucorales; family, Mucoraceae; genus, Mucor; species, miehei; variety Cooney et Emerson.

(b) The strain of Mucor miehei var. Cooney et Emerson is nonpathogenic and nontoxic in man or other animals.
§ 173.145 Alpha-Galactosidase derived from Mortierella vinaceae var. raffinosoavorant.

The food additive alpha-galactosidase and parent mycelial microorganism Mortierella vinaceae var. raffinosoavorant may be safely used in food in accordance with the following conditions:

(a) The food additive is the enzyme alpha-galactosidase and the mycelia of the microorganism Mortierella vinaceae var. raffinosoavorant which produces the enzyme.

(b) The nonpathogenic microorganism matches American Type Culture Collection (ATCC) No. 20034,1 and is classified as follows:

Class: Phycocystes.
Order: Mucorales.
Family: Mortierellaceae.
Genus: Mortierella.
Species: vinaceae.
Variety: raffinosoavorant.

(c) The additive is used or intended for use in the production of sugar (sucrose) from sugar beets by addition of mycelial pellets to the molasses to increase the yield of sucrose, followed by removal of the spent mycelial pellets by filtration.

1Available from: American Type Culture Collection, 12301 Parklawn Drive, Rockville, MD 20852.
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§ 173.160 Candida guilliermondii.

The food additive Candida guilliermondii may be safely used as the organism for fermentation production of citric acid in accordance with the following conditions:

(a) The food additive is the enzyme system of the viable organism Candida guilliermondii and its concomitant metabolites produced during the fermentation process.

(b)(1) The nonpathogenic and non-toxicogenic organism descending from strain, American Type Culture Collection (ATCC) No. 20474, is classified as follows:

Class: Deuteromycetes.
Order: Moniliales.
Family: Cryptococcaceae.
Genus: Candida.
Species: guilliermondii.
Variety: guilliermondii.

(2) The taxonomic characteristics of the reference culture strain ATCC No. 20474 agree in the essentials with the standard description for Candida guilliermondii variety guilliermondii listed in "The Yeasts—A Taxonomic Study:" 2d Ed. (1970), by J acomina Lodder, which is incorporated by reference. Copies are available from the Center for Food Safety and Applied Nutrition (HFS-200), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

(c)(1) The additive is used or intended for use as a pure culture in the fermentation process for the production of citric acid using an acceptable aqueous carbohydrate substrate.

(c)(2) The organism Candida guilliermondii is made nonviable and is completely removed from the citric acid during the recovery and purification process.

(d) The additive is so used that the citric acid produced conforms to the specifications of the "Food Chemicals Codex," 3d Ed. (1981), under "Citric acid," pp. 86-87, which is incorporated by reference. Copies may be obtained from the National Academy Press, 2101 Constitution Ave. NW., Washington, DC 20418, or may be examined at the Office of the Federal Register., 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

§ 173.165 Candida lipolytica.

The food additive Candida lipolytica may be safely used as the organism for fermentation production of citric acid in accordance with the following conditions:

(a) The food additive is the enzyme system of the organism Candida lipolytica and its concomitant metabolites produced during the fermentation process.

(b)(1) The nonpathogenic organism is classified as follows:

Class: Deuteromycetes.
Order: Moniliales.
Family: Cryptococcaceae.
Genus: Candida.
Species: lipolytica.
Variety: lipolytica.

(2) The taxonomic characteristics of the culture agree in essential with the standard description for Candida lipolytica variety lipolytica listed in "The Yeasts—A Taxonomic Study:" 2d Ed. (1970), by J acomina Lodder, which is incorporated by reference. Copies are available from the Center for Food Safety and Applied Nutrition (HFS-200), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

(c) The additive is used or intended for use as a pure culture in the fermentation process for the production of citric acid from purified normal alkanes.

(d) The additive is so used that the citric acid produced conforms to the specifications of the "Food Chemicals Codex," 3d Ed. (1981), pp. 86-87, which is incorporated by reference. Copies may

1Available from: American Type Culture Collection, 12301 Parklawn Drive, Rockville, MD 20852.
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be obtained from the National Academy Press, 2101 Constitution Ave. NW., Washington, DC 20418, or may be examined at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408. The additive meets the following ultraviolet absorbance limits when subjected to the analytical procedure described in this paragraph:

<table>
<thead>
<tr>
<th>Ultraviolet absorbance per centimeter path length</th>
<th>Maximum</th>
</tr>
</thead>
<tbody>
<tr>
<td>280 to 289 millimicrons</td>
<td>0.25</td>
</tr>
<tr>
<td>290 to 299 millimicrons</td>
<td>0.20</td>
</tr>
<tr>
<td>300 to 309 millimicrons</td>
<td>0.13</td>
</tr>
<tr>
<td>310 to 400 millimicrons</td>
<td>0.03</td>
</tr>
</tbody>
</table>

ANALYTICAL PROCEDURE FOR CITRIC ACID

GENERAL INSTRUCTIONS

Because of the sensitivity of the test, the possibility of errors arising from contamination is great. It is of the greatest importance that all glassware be scrupulously cleaned to remove all organic matter such as oil, grease, detergent residues, etc. Examine all glassware including stoppers and stopcocks, under ultraviolet light to detect any residual fluorescent contamination. As a precautionary measure it is recommended practice to rinse all glassware with purified isooctane immediately before use. No grease is to be used on stopcocks or joints. Great care to avoid contamination of citric acid samples in handling is essential to assure absence of any extraneous material arising from inadequate packaging. Because some of the polynuclear hydrocarbons sought in this test are very susceptible to photo-oxidation, the entire procedure is to be carried out under subdued light.

APPARATUS

1. Aluminum foil, oil free.
2. Separatory funnels, 500-milliliter capacity, equipped with tetrafluoroethylene polymer stopcocks.
3. Chromatographic tubes: (a) 30-milliliter ID × 900-millimeter length equipped with tetrafluoroethylene polymer stopcock and course fritted disk; (b) 18-milliliter ID × 300-millimeter length equipped with tetrafluoroethylene polymer stopcock.
4. Rotary vacuum evaporator, Buchi or equivalent.
5. Spectrophotometer—Spectral range 250-400 nanometers with spectral slit width of 2 nanometers or less; under instrument operating conditions for these absorbance measurements, the spectrophotometer shall also meet the following performance requirements:
   - Absorbance repeatability, ±0.01 at 0.4 absorbance.
   - Wavelength repeatability, ±0.2 nanometer.
   - Wavelength accuracy, ±0.1 nanometer.

The spectrophotometer is equipped with matched 1 centimeter path length quartz microcuvettes with 0.5-milliliter volume capacity.

6. Vacuum oven, minimum inside dimensions: 200 mm × 200 mm × 300 mm deep.

REAGENTS AND MATERIALS

Organic solvents. All solvents used throughout the procedure shall meet the specifications and tests described in this specification. The methyl alcohol, isooctane, benzene, hexane and 1,2-dichloroethane designated in the list following this paragraph shall pass the following test:

The specified quantity of solvent is added to a 250-milliliter round bottom flask containing 0.5 milliliter of purified n-hexadecane and evaporated on the rotary evaporator at 45 ºC to constant volume. Six milliliters of purified isooctane are added to this residue and evaporated under the same conditions as above for 5 minutes. Determine the absorbance of the residue compared to purified n-hexadecane as reference. The absorbance of the solution of the solvent residue shall not exceed 0.03 per centimeter path length between 280 and 299 nanometers and 0.05 per centimeter path length between 300 and 400 nanometers.

Methyl alcohol, A.C.S. reagent grade. Use 100 milliliters for the test described in the preceding paragraph. If necessary, methyl alcohol may be purified by distillation through a Virgreaux column discarding the first and last ten percent of the distillate or otherwise.

Benzene, spectrograde (Burdick and Jackson Laboratories, Inc., Muskegon, Mich., or equivalent). Use 80 milliliters for the test. If necessary, benzene may be purified by distillation or otherwise.

Isooctane (2,2,4-trimethylpentane). Use 100 milliliters for the test. If necessary, isooctane may be purified by passage through a column of activated silica gel, distillation or otherwise.

Hexane, spectrograde (Burdick and Jackson Laboratories, Inc., Muskegon, Mich., or equivalent). Use 100 milliliters for the test. If necessary, hexane may be purified by distillation or otherwise.

1,2-Dichloroethane, spectrograde (Matheson, Coleman and Bell, East Rutherford, N.J., or equivalent). Use 100 milliliters for the test. If necessary, 1,2-dichloroethane may be purified by distillation or otherwise.

ELUTING MIXTURES

1. 10 percent 1,2-dichloroethane in hexane. Prepare by mixing the purified solvents in the volume ratio of 1 part of 1,2-dichloroethane to 9 parts of hexane.
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2. 40 percent benzene in hexane. Prepare by mixing the purified solvents in the volume ratio of 4 parts of benzene to 6 parts of hexane.

n-Hexadecane, 99 percent olefin-free. Determine the absorbance compared to isoctane as reference. The absorbance per centimeter path length shall not exceed 0.00 in the range of 283-400 nanometers. If necessary, n-hexadecane may be purified by percolation through activated silica gel, distillation or otherwise.

Silica gel, 28-200 mesh (Grade 12, Davison Chemical Co., Baltimore, Md., or equivalent). Activate as follows: Slurry 900 grams of silica gel reagent with 2 liters of purified water in a 3 liter beaker. Cool the mixture and pour into a 80 x 900 chromatographic column with coarse fritted disc. Drain the water, wash with an additional 6 liters of purified water and wash with 3,600 milliliters of purified methyl alcohol at a relatively slow rate. Drain all of the solvents and transfer the silica gel to an aluminum foil-lined drying dish. Place foil over the top of the dish. Activate in a vacuum oven at low vacuum (approximately 750 millimeters Mercury or 27 inches of Mercury below atmospheric pressure) at 173° to 177° C for at least 20 hours. Cool under vacuum and store in an amber bottle.

Sodium sulfate, anhydrous, A.C.S. reagent grade. This reagent should be washed with purified isoctane. Check the purity of this reagent as described in §172.86 of this chapter.

Water, purified. All water used must meet the specifications of the following test:

Extract 600 milliliters of water with 50 milliliters of purified water. Add 1 milliliter of purified n-hexadecane to the isoctane extract and evaporate the resulting solution to 1 milliliter. The absorbance of this residue shall not exceed 0.02 per centimeter path length between 300-400 nanometers and 0.03 per centimeter path length between 280-299 nanometers. If necessary, water may be purified by distillation, extraction with purified organic solvents, treatment with an absorbent (e.g., activated carbon) followed by filtration of the absorbent if otherwise.

PROCEDURE

Separate portions of 200 milliliters of purified water are taken through the procedure for use as control blanks. Each citric acid sample is processed as follows: Weigh 200 grams of anhydrous citric acid into a 500 milliliter flask and dissolve in 200 milliliters of pure water. Heat the solution to 60° C and transfer to a 500 milliliter separatory funnel. Rinse the flask with 50 milliliters of isoctane and add the isoctane to the separatory funnel. Gently shake the mixture 90 times (caution: vigorous shaking will cause emulsions) with periodic release of the pressure caused by shaking.

Allow the phases to separate for at least 5 minutes. Draw off the lower aqueous layer into a second 500-milliliter separatory funnel and repeat the extraction with a second aliquot of 50 milliliters of isoctane. After separation of the layers, draw off and discard the water layer. Combine both isoctane extracts in the funnel containing the first extract. Rinse the funnel which contained the second extract with 10 milliliters of isoctane and add this portion to the combined isoctane extract.

A chromatographic column containing 5.5 grams of silica gel and 3 grams of anhydrous sodium sulfate is prepared for each citric acid sample as follows: Fit 18 x 300 column with a small glass wool plug. Rinse the inside of the column with 30 milliliters of purified isoctane. Drain the isoctane from the column. Pour 5.5 grams of activated silica gel into the column. Tap the column approximately 20 times on a semisoft, clean surface to settle the silica gel. Carefully pour 3 grams of anhydrous sodium sulfate onto the top of the silica gel in the column. Carefully drain the isoctane extract of the citric acid solution into the column in a series of additions while the isoctane is draining from the column at an elution rate of approximately 3 milliliters per minute. Rinse the separatory funnel with 10 milliliters of isoctane after the last portion of the extract has been applied to the column and add this rinse to the column. After all of the extract has been applied to the column and the solvent layer reaches the top of the sulfate bed, rinse the column with 25 milliliters of isoctane followed by 10 milliliters of a 10-percent dichloromethane in hexane solution. For each rinse solution, drain the column until the solvent layer reaches the top of the sulfate bed. Discard the rinse solvents. Place a 250-milliliter round bottom flask containing 0.5 milliliter of purified n-hexadecane under the column. Elute the polynuclear aromatic hydrocarbons from the column with 30 milliliters of 40-percent benzene in hexane solution. Drain the eluate until the 40-percent benzene in the hexane solvent reaches the top of the sulfate bed.

Evaporate the 40-percent benzene in hexane eluate on the rotary vacuum evaporator at 45° C until only the n-hexadecane residue of 0.5 milliliter remains. Treat the n-hexadecane residue twice with the following wash step: Add 6 milliliters of purified isoctane and remove the solvents by vacuum evaporation at 45° C to constant volume, i.e., 0.5 milliliter. Cool the n-hexadecane residue and transfer the solution to an 0.5-milliliter microcuvette. Determine the absorbance of this solution compared to purified n-hexadecane as reference. Correct the absorbance values for any absorbance derived from the
control reagent blank. If the corrected absorbance does not exceed the limits prescribed, the samples meet the ultraviolet absorbance specifications.

The reagent blank is prepared by using 200 milliliters of purified water in place of the citric acid solution and carrying the water sample through the procedure. The typical control reagent blank should not exceed 0.03 absorbance per centimeter path length between 280 and 299 nanometers, 0.02 absorbance per centimeter path length between 300 and 359 nanometers, and 0.01 absorbance per centimeter path length between 360 and 400 nanometers.


§ 173.170 Aminoglycoside 3'-phosphotransferase II.

The food additive aminoglycoside 3'-phosphotransferase II may be safely used in the development of genetically modified cotton, oilseed rape, and tomatoes in accordance with the following prescribed conditions:

(a) The food additive is the enzyme aminoglycoside 3'-phosphotransferase II (CAS Reg. No. 58943-39-8) which catalyzes the phosphorylation of certain aminoglycoside antibiotics, including kanamycin, neomycin, and gentamicin.

(b) Aminoglycoside 3'-phosphotransferase II is encoded by the kan^r gene originally isolated from transposon Tn^3 of the bacterium Escherichia coli.

(c) The level of the additive does not exceed the amount reasonably required for selection of plant cells carrying the kan^r gene along with the genetic material of interest.

[59 FR 26711, May 23, 1994]

Subpart C—Solvents, Lubricants, Release Agents and Related Substances

§ 173.210 Acetone.

A tolerance of 30 parts per million is established for acetone in spice oleoresins when present therein as a residue from the extraction of spice.

§ 173.220 1,3-Butylene glycol.

1,3-Butylene glycol (1,3-butanediol) may be safely used in food in accordance with the following prescribed conditions:

(a) The substance meets the following specifications:

(1) 1,3-Butylene glycol content: Not less than 99 percent.

(2) Specific gravity at 20°/20°C: 1.004 to 1.006.

(3) Distillation range: 200°–215°C.

(b) It is used in the minimum amount required to perform its intended effect.

(c) It is used as a solvent for natural and synthetic flavoring substances except where standards of identity issued under section 401 of the act preclude such use.

§ 173.228 Ethyl acetate.

Ethyl acetate (CAS Reg. No. 141-78-6) may be safely used in food in accordance with the following conditions:

(a) The additive meets the specifications of the Food Chemicals Codex, 1 (Ethyl Acetate, p. 372, 3d Ed., 1981), which are incorporated by reference.

(b) The additive is used in accordance with current good manufacturing practice as a solvent in the decaffeination of coffee and tea.

[47 FR 146, Jan. 5, 1982, as amended at 49 FR 28548, July 13, 1984]

§ 173.230 Ethylene dichloride.

A tolerance of 30 parts per million is established for ethylene dichloride in spice oleoresins when present therein as a residue from the extraction of spice; Provided, however, That if residues of other chlorinated solvents are also present the total of all residues of such solvents shall not exceed 30 parts per million.

§ 173.240 Isopropyl alcohol.

Isopropyl alcohol may be present in the following foods under the conditions specified:

(a) In spice oleoresins as a residue from the extraction of spice, at a level not to exceed 50 parts per million.

(b) In lemon oil as a residue in production of the oil, at a level not to exceed 6 parts per million.

1Copies may be obtained from: National Academy Press, 2101 Constitution Ave. NW., Washington, DC 20418 or examined at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.
Food and Drug Administration, HHS

§ 173.280 Solvent extraction process for citric acid.

A solvent extraction process for recovery of citric acid from conventional Aspergillus niger fermentation liquor may be safely used to produce food-grade citric acid in accordance with the following conditions:

(a) The solvent used in the process consists of a mixture of n-octyl alcohol meeting the requirements of §172.864 of this chapter, synthetic isoparaffinic petroleum hydrocarbons meeting the requirements of §172.882 of this chapter, and tridodecyl amine.

(b) In hops extract as a residue from the extraction of hops at a level not to exceed 2.0 percent by weight: Provided, That,

(1) The hops extract is added to the wort before or during cooking in the manufacture of beer.

(2) The label of the hops extract specifies the presence of the isopropyl alcohol and provides for the use of the hops extract only as prescribed by paragraph (c)(1) of this section.

§ 173.250 Methyl alcohol residues.

Methyl alcohol may be present in the following foods under the conditions specified:

(a) In spice oleoresins as a residue from the extraction of spice, at a level not to exceed 50 parts per million.

(b) In hops extract as a residue from the extraction of hops, at a level not to exceed 2.2 percent by weight; Provided, That:

(1) The hops extract is added to the wort before or during cooking in the manufacture of beer.

(2) The label of the hops extract specifies the presence of the methyl alcohol and provides for the use of the hops extract only as prescribed by paragraph (b)(1) of this section.

§ 173.255 Methylene chloride.

Methylene chloride may be present in food under the following conditions:

(a) In spice oleoresins as a residue from the extraction of spice, at a level not to exceed 30 parts per million; Provided, That, if residues of other chlorinated solvents are also present, the total of all residues of such solvents shall not exceed 30 parts per million.

(b) In hops extract as a residue from the extraction of hops, at a level not to exceed 2.2 percent, Provided, That:

(1) The hops extract is added to the wort before or during cooking in the manufacture of beer.

(2) The label of the hops extract specifies the presence of methyl alcohol and provides for the use of the hops extract only as prescribed by paragraph (b)(1) of this section.

§ 173.270 Hexane.

Hexane may be present in the following foods under the conditions specified:

(a) In spice oleoresins as a residue from the extraction of spice, at a level not to exceed 25 parts per million.

(b) In hops extract as a residue from the extraction of hops, at a level not to exceed 2.2 percent by weight; Provided, That:

(1) The hops extract is added to the wort before or during cooking in the manufacture of beer.

(2) The label of the hops extract specifies the presence of the hexane and provides for the use of the hops extract only as prescribed by paragraph (b)(1) of this section.

§ 173.275 Hydrogenated sperm oil.

The food additive hydrogenated sperm oil may be safely used in accordance with the following prescribed conditions:

(a) The sperm oil is derived from rendering the fatty tissue of the sperm whale or is prepared by synthesis of fatty acids and fatty alcohols derived from the sperm whale. The sperm oil obtained by rendering is refined. The oil is hydrogenated.

(b) It is used alone or as a component of a release agent or lubricant in bakery pans.

(c) The amount used does not exceed that reasonably required to accomplish the intended lubricating effect.

§ 173.280 Solvent extraction process for citric acid.

A solvent extraction process for recovery of citric acid from conventional Aspergillus niger fermentation liquor may be safely used to produce food-grade citric acid in accordance with the following conditions:

(a) The solvent used in the process consists of a mixture of n-octyl alcohol meeting the requirements of §172.864 of this chapter, synthetic isoparaffinic petroleum hydrocarbons meeting the requirements of §172.882 of this chapter, and tridodecyl amine.

percent) in decaffeinated roasted coffee and in decaffeinated soluble coffee extract (instant coffee).
§ 173.290  Trichloroethylene.

(a) The additive is generated by treating an aqueous solution of sodium chlorite with either chlorine gas or a mixture of sodium hypochlorite and hydrochloric acid. The generator effluent contains at least 90 percent (by weight) of chlorine dioxide with respect to all chlorine species as determined by Method 4500-C1O2 E in the "Standard Methods for the Examination of Water and Wastewater," 18th ed., 1992, or an equivalent method. Method 4500-C1O2 E is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the Center for Food Safety and Applied Nutrition (HFS-200), Food And Drug Administration, 200 C St., SW., Washington, DC 20204-0003 and The American Public Health Association, 1015 Fifteenth St., NW., Washington, DC 20005, or may be examined at the Office of the Federal Register, 800 North Capitol St., NW., suite 700, Washington, DC.

(b)(1) The additive may be used as an antimicrobial agent in water used in poultry processing in an amount not to exceed 3 parts per million (ppm) residual chlorine dioxide as determined by Method 4500-C1O2 E, referenced in paragraph (a) of this section, or an equivalent method.

(b)(2) The additive may be used as an antimicrobial agent in water used to wash fruits and vegetables that are not raw agricultural commodities in an amount not to exceed 3 ppm residual chlorine dioxide as determined by Method 4500-C1O2 E, referenced in paragraph (a) of this section, or an equivalent method. Treatment of the fruits and vegetables with chlorine dioxide shall be followed by a potable water rinse or by blanching, cooking, or canning.

§ 173.310  Boiler water additives.

Boiler water additives may be safely used in the preparation of steam that will contact food, under the following conditions:

(a) The amount of additive is not in excess of that required for its functional purpose, and the amount of
steams in contact with food does not exceed that required to produce the intended effect in or on the food.

(b) The compounds are prepared from substances identified in paragraphs (c) and (d) of this section, and are subject to the limitations, if any, prescribed:

(c) List of substances:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Acrylamide-sodium acrylate resin</td>
<td>Contains not more than 0.05 percent by weight of acrylamide monomer.</td>
</tr>
<tr>
<td>Acrylic acid/2-acrylamido-2-methyl propane sulfonic acid copolymer</td>
<td>Total not to exceed 20 parts per million (active) in boiler feedwater.</td>
</tr>
<tr>
<td>having a minimum weight average molecular weight of 9,900 and a minimum</td>
<td></td>
</tr>
<tr>
<td>number average molecular weight of 5,700 as determined by a method</td>
<td></td>
</tr>
<tr>
<td>entitled “Determination of Weight Average and Number Average Molecular</td>
<td></td>
</tr>
<tr>
<td>Weight of 60/40 AA/AMPS” (October 23, 1987), which is incorporated</td>
<td></td>
</tr>
<tr>
<td>by reference in accordance with 5 U.S.C. 552(a). Copies may</td>
<td></td>
</tr>
<tr>
<td>be obtained from the Center for Food Safety and Applied Nutrition (HFS-200), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or may be examined at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC.</td>
<td></td>
</tr>
<tr>
<td>Ammonium alginate,</td>
<td></td>
</tr>
<tr>
<td>Cobalt sulfate (as catalyst).</td>
<td></td>
</tr>
<tr>
<td>1-hydroxyethylidene-1,1-diphosphonic acid (CAS Reg. No. 26099-09-2), and</td>
<td></td>
</tr>
<tr>
<td>its sodium and potassium salts.</td>
<td></td>
</tr>
<tr>
<td>Lignosulfonic acid.</td>
<td></td>
</tr>
<tr>
<td>Monobutyl ethers of polyethylene-polypropylene glycol produced by</td>
<td></td>
</tr>
<tr>
<td>random condensation of a 1:1 mixture by weight of ethylene oxide and</td>
<td></td>
</tr>
<tr>
<td>propylene oxide with butanol.</td>
<td></td>
</tr>
<tr>
<td>Poly(acrylic acid-co-hypophosphite), sodium salt (CAS Reg. No. 71050-62-9), produced from a 4:1 to a 16:1 mixture by weight of acrylic acid and sodium hypophosphite.</td>
<td></td>
</tr>
<tr>
<td>Polyethylene glycol.</td>
<td>Minimum mol. wt. 1,500.</td>
</tr>
<tr>
<td>Polyoxypropylene glycol</td>
<td>Total not to exceed 1.5 parts per million in boiler feed water.</td>
</tr>
<tr>
<td>Potassium carbonate.</td>
<td>Copolymer contains not more than 0.5 percent by weight of acrylic acid monomer (dry weight basis).</td>
</tr>
<tr>
<td>Potassium tripolyphosphate.</td>
<td>As defined in §172.620 of this chapter.</td>
</tr>
<tr>
<td>Sodium acetate.</td>
<td>Total not to exceed 1 part per million in boiler feed water (calculated as the acid).</td>
</tr>
<tr>
<td>Sodium alginate.</td>
<td>Minimum mol. wt. 1,000.</td>
</tr>
<tr>
<td>Sodium aluminate.</td>
<td></td>
</tr>
<tr>
<td>Sodium carbonate.</td>
<td></td>
</tr>
<tr>
<td>Sodium carboxymethylcellulose</td>
<td></td>
</tr>
<tr>
<td>Sodium glucoheptonate.</td>
<td>Contains not less than 95 percent sodium carboxymethylcellulose on a dry-weight basis, with maximum substitution of 0.9 carboxymethylcellulose groups per anhydroglucose unit, and with a minimum viscosity of 15 centipoises for 2 percent by weight aqueous solution at 25 °C, by the method prescribed in the “Food Chemicals Codex,” 4th ed. (1996), pp. 744–745, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the National Academy Press, Box 285, 2101 Constitution Ave. NW., Washington, DC 20055 (Internet address “<a href="http://www.nap.edu%E2%80%9D">http://www.nap.edu”</a>), or may be examined at the Center for Food Safety and Applied Nutrition’s Library, Food and Drug Administration, 200 C St. SW., rm. 3321, Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC.</td>
</tr>
<tr>
<td>Sodium hexametaphosphate.</td>
<td>Less than 1 part per million cyanide in the sodium glucoheptonate.</td>
</tr>
<tr>
<td>Sodium humate.</td>
<td></td>
</tr>
<tr>
<td>Sodium hydroxide.</td>
<td></td>
</tr>
<tr>
<td>Sodium lignosulfonate.</td>
<td></td>
</tr>
<tr>
<td>Sodium metabisulfite.</td>
<td></td>
</tr>
<tr>
<td>Sodium metasilicate.</td>
<td></td>
</tr>
<tr>
<td>Sodium nitrate.</td>
<td></td>
</tr>
<tr>
<td>Sodium phosphate (mono-, di-, tri-).</td>
<td></td>
</tr>
<tr>
<td>Sodium polyacrylate.</td>
<td></td>
</tr>
<tr>
<td>Sodium polymethacrylate.</td>
<td></td>
</tr>
<tr>
<td>Sodium silicate.</td>
<td></td>
</tr>
<tr>
<td>Sodium sulfate.</td>
<td></td>
</tr>
<tr>
<td>Sodium sulfite (neutral or alkaline).</td>
<td></td>
</tr>
</tbody>
</table>
§ 173.315

Substances Limitations

Sodium tripolyphosphate. The mixture is used as an antirust agent in steam boiler distribution systems, with each component not to exceed 15 parts per million in the steam.

Sorbitol anhydride esters: a mixture consisting of sorbitan monostearate as defined in § 172.842 of this chapter; polysorbate 60 ((polyoxyethylene (20) sorbitan monostearate)) as defined in § 172.836 of this chapter; and polysorbate 20 ((polyoxyethylene (20) sorbitan monolaurate)), meeting the specifications of the Food Chemicals Codex, 4th ed. (1996), pp. 306–307, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the National Academy Press, 2101 Constitution Ave. NW., Box 285, Washington, DC 20055 (Internet http://www.nap.edu), or may be examined at the Center for Food Safety and Applied Nutrition’s Library, Food and Drug Administration, 200 C St. SW., rm. 3321, Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC. Tannin (including quebracho extract). Tetrasodium EDTA. Tetrasodium pyrophosphate.

(d) Substances used alone or in combination with substances in paragraph (c) of this section:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cyclohexylamine</td>
<td>Not to exceed 10 parts per million in steam, and excluding use of such steam in contact with milk and milk products.</td>
</tr>
<tr>
<td>Diethylaminoethanol</td>
<td>Not to exceed 15 parts per million in steam, and excluding use of such steam in contact with milk and milk products.</td>
</tr>
<tr>
<td>Hydrazine</td>
<td>Zero in steam.</td>
</tr>
<tr>
<td>Morpholine</td>
<td>Not to exceed 10 parts per million in steam, and excluding use of such steam in contact with milk and milk products.</td>
</tr>
<tr>
<td>Octadecylamine</td>
<td>Not to exceed 3 parts per million in steam, and excluding use of such steam in contact with milk and milk products.</td>
</tr>
<tr>
<td>Trisodium nitrilotriacetate</td>
<td>Not to exceed 5 parts per million in boiler feedwater; not to be used where steam will be in contact with milk and milk products.</td>
</tr>
</tbody>
</table>

(e) To assure safe use of the additive, in addition to the other information required by the Act, the label or labeling shall bear:

(1) The common or chemical name or names of the additive or additives.

(2) Adequate directions for use to assure compliance with all the provisions of this section.

Food and Drug Administration, HHS

§ 173.315

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>A mixture of alkylene oxide adducts of alkyl alcohols and phosphate esters of alkylene oxide adducts of alkyl alcohols consisting of: α-alkyl (C₁₂-C₁₈)-ω-hydroxy-poly(oxyethylene) (7.5–8.5 moles/poly(oxypropylene)) block copolymer having an average molecular weight of 810; ω-alkyl (C₁₂-C₁₈)-ω-hydroxy-poly(oxyethylene) (3.3–3.7 moles/poly(oxypropylene)) polymer having an average molecular weight of 380, and subsequently esterified with 1.25 moles phosphonic anhydride; and ω-alkyl (C₂₂-C₃₄)-ω-hydroxy-poly(oxyethylene) (11.9–12.9 moles/poly(oxypropylene)) copolymer, having an average molecular weight of 610, and subsequently esterified with 1.25 moles phosphonic anhydride. Aliphatic acid mixture consisting of capric, caprylic, caprylic, and pelargonic acids. Polycrylamide</td>
<td>May be used at a level not to exceed 0.2 percent in lye-peeling solution to assist in the lye peeling of fruit and vegetables.</td>
</tr>
<tr>
<td>Potassium bromide</td>
<td></td>
</tr>
<tr>
<td>Sodium N-alkylbenzene-sulfonate (alkyl group predominantly C₁₂ and not less than 95 percent C₁₂ to C₁₈).</td>
<td>May be used in the washing or to assist in the lye peeling of fruits and vegetables.</td>
</tr>
<tr>
<td>Sodium dodecylbenzene-sulfonate (alkyl group predominantly C₁₂ and not less than 95 percent C₁₂ to C₁₈).</td>
<td>Do.</td>
</tr>
<tr>
<td>Sodium 2 ethylhexyl sulfate</td>
<td>Do.</td>
</tr>
<tr>
<td>Sodium hypochlorite</td>
<td>May be used in the washing or to assist in the lye peeling of fruits and vegetables.</td>
</tr>
<tr>
<td>Sodium mono- and dimethyl naphthalene sulfonates (mol. wt. 245–260)</td>
<td>Not to exceed 0.2 percent in wash water. May be used in the washing or to assist in the lye peeling of fruit and vegetables.</td>
</tr>
</tbody>
</table>

(3) Sodium mono- and dimethyl naphthalene sulfonates (mol. wt. 245–260) may be used in the steam/scald vacuum peeling of tomatoes at a level not to exceed 0.2 percent in the condensate or scald water.

(4) Substances identified in this paragraph (a)(4) for use in flume water for washing sugar beets prior to the slicing operation and subject to the limitations as are provided for the level of the substances in the flume water:

<table>
<thead>
<tr>
<th>Substance</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>α-Alkyl-ω-hydroxy-poly(oxyethylene) produced by condensation of 1 mole of C₁₂-C₄₈/G₁₂-H₃₄) straight chain randomly substituted secondary alcohols with an average of 9 moles of ethylene oxide. Linear undecylbenzenesulfonic acid. Diethanolamide produced by condensing 1 mole of methyl laurate with 1.05 moles of dethanolamine. Triethanolamine</td>
<td>Not to exceed 3 ppm.</td>
</tr>
<tr>
<td>Oleic acid conforming with § 172.860 of this chapter. Peroxyacetic acid</td>
<td>Not to exceed 2 ppm.</td>
</tr>
<tr>
<td>1-Hydroxyethylidene-1,1-diphosphonic acid. Peroxycetic acid</td>
<td>Not to exceed 1 ppm.</td>
</tr>
<tr>
<td>Do.</td>
<td>Not to exceed 0.3 ppm.</td>
</tr>
<tr>
<td>Do.</td>
<td>Not to exceed 0.2 ppm.</td>
</tr>
<tr>
<td>Do.</td>
<td>Not to exceed 0.1 ppm.</td>
</tr>
</tbody>
</table>

(5) Substances identified in this paragraph (a)(5) for use on fruits and vegetables that are not raw agricultural commodities and subject to the limitations provided:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Hydrogen peroxide</td>
<td>Used in combination with acetic acid to form peroxyacetic acid. Not to exceed 59 ppm in wash water.</td>
</tr>
<tr>
<td>1-Hydroxyethylidene-1,1-diphosphonic acid. Peroxycetic acid</td>
<td>Prepared by reacting acetic acid with hydrogen peroxide. Not to exceed 80 ppm in wash water.</td>
</tr>
</tbody>
</table>

(b) The chemicals are used in amounts not in excess of the minimum required to accomplish their intended effect.

(c) The use of the chemicals listed under paragraphs (a)(1), (a)(2), and (a)(4) is followed by rinsing with potable water to remove, to the extent possible, residues of the chemicals.

(d) To assure safe use of the additive:

(1) The label and labeling of the additive container shall bear, in addition to the other information required by the
§ 173.320 Chemicals for controlling microorganisms in cane-sugar and beet-sugar mills.

Agents for controlling microorganisms in cane-sugar and beet-sugar mills may be safely used in accordance with the following conditions:

(a) They are used in the control of microorganisms in cane-sugar and/or beet-sugar mills as specified in paragraph (b) of this section.

(b) They are applied to the sugar mill grinding, crusher, and/or diffuser systems in one of the combinations listed in paragraph (b) (1), (2), (3), or (5) of this section or as a single agent listed in paragraph (b) (4) or (6) of this section. Quantities of the individual additives in parts per million are expressed in terms of the weight of the raw cane or raw beets.

(1) Combination for cane-sugar mills:

<table>
<thead>
<tr>
<th>Parts per million</th>
</tr>
</thead>
<tbody>
<tr>
<td>Disodium cyanodithioimidocarbonate</td>
</tr>
<tr>
<td>Ethylenediamine</td>
</tr>
<tr>
<td>Potassium $N$-methylthiocarbamate</td>
</tr>
</tbody>
</table>

(2) Combination for cane-sugar mills:

<table>
<thead>
<tr>
<th>Parts per million</th>
</tr>
</thead>
<tbody>
<tr>
<td>Disodium ethylenebisdithiocarbamate</td>
</tr>
<tr>
<td>Sodium dimethylthiocarbamate</td>
</tr>
</tbody>
</table>

(3) Combinations for cane-sugar mills and beet-sugar mills:

<table>
<thead>
<tr>
<th>Parts per million</th>
</tr>
</thead>
<tbody>
<tr>
<td>(i) Disodium ethylenebisdithiocarbamate</td>
</tr>
<tr>
<td>Ethylenediamine</td>
</tr>
<tr>
<td>Sodium dimethylthiocarbamate</td>
</tr>
<tr>
<td>(ii) Disodium cyanodithioimidocarbonate</td>
</tr>
<tr>
<td>Potassium $N$-methylthiocarbamate</td>
</tr>
</tbody>
</table>

(4) Single additive for cane-sugar mills and beet-sugar mills.

<table>
<thead>
<tr>
<th>Parts per million</th>
</tr>
</thead>
<tbody>
<tr>
<td>2,2-Dibromo-3-nitropropionamide (CAS Reg. No. 10222–01–2). Limitations: By-product molasses, bagasse, and pulp containing residues of 2,2-dibromo-3-nitropropionamide are not authorized for use in animal feed.</td>
</tr>
</tbody>
</table>

| Not more than 10.0 and not less than 2.0. |

(5) Combination for cane-sugar mills:

<table>
<thead>
<tr>
<th>Parts per million</th>
</tr>
</thead>
<tbody>
<tr>
<td>$n$-Dodecyl dimethyl benzyl ammonium chloride</td>
</tr>
<tr>
<td>$n$-Dodecyl dimethyl ethylbenzyl ammonium chloride</td>
</tr>
<tr>
<td>$n$-Hexadecyl dimethyl benzyl ammonium chloride</td>
</tr>
<tr>
<td>$n$-Octadecyl dimethyl benzyl ammonium chloride</td>
</tr>
<tr>
<td>$n$-Tetradecyl dimethyl benzyl ammonium chloride</td>
</tr>
<tr>
<td>$n$-Tetradecyl dimethyl ethylbenzyl ammonium chloride</td>
</tr>
</tbody>
</table>

Limitations. Byproduct molasses, bagasse, and pulp containing residues of these quaternary ammonium salts are not authorized for use in animal feed.

(6) Single additive for beet-sugar mills:

<table>
<thead>
<tr>
<th>Parts per million</th>
</tr>
</thead>
<tbody>
<tr>
<td>Glutaraldehyde (CAS Reg. No. 111–30–8). Not more than 250.</td>
</tr>
</tbody>
</table>

(c) To assure safe use of the additives, their label and labeling shall conform to that registered with the Environmental Protection Agency.


§ 173.322 Chemicals used in delinting cottonseed.

Chemicals may be safely used to assist in the delinting of cottonseed in accordance with the following conditions:

(a) The chemicals consist of one or more of the following:

(1) Substances generally recognized as safe for direct addition to food.

(2) Substances identified in this paragraph and subject to such limitations as are provided:

Not more than 250.

VerDate 27<APR>2000 14:28 May 10, 2000 Jkt 190063 PO 00000 Frm 00132 Fmt 8010 Sfmt 8010 Y:\SGML\190063T.XXX pfrm02 PsN: 190063T
Substances

alpha-Alkyl-omega-hydroxy-polyoxyethylene produced by condensation of a linear primary alcohol containing an average chain length of 10 carbons with poly(oxyethylene) having an average of 5 ethylene oxide units.


Limitations

May be used at an application rate not to exceed 0.3 percent by weight of cottonseeds to enhance delinting of cottonseeds intended for the production of cottonseed oil. Byproducts including lint, hulls, and meal may be used in animal feed.

Acidified sodium chlorite solutions may be safely used in accordance with the following prescribed conditions:

(a) The additive is produced by mixing an aqueous solution of sodium chlorite (CAS Reg. No. 7758-19-2) with any generally recognized as safe (GRAS) acid.

(b)(1) The additive is used as an antimicrobial agent in poultry processing water in accordance with current industry practice under the following conditions:

(i) As a component of a carcass spray or dip solution prior to immersion of the intact carcass in a prechiller or chiller tank;

(ii) In a prechiller or chiller solution for application to the intact carcass;

(iii) As a component of a spray or dip solution for application to poultry carcass parts; or

(iv) In a prechiller or chiller solution for application to poultry carcass parts.

(2) When used in a spray or dip solution, the additive is used at levels that result in sodium chlorite concentrations between 500 and 1,200 parts per million (ppm), in combination with any GRAS acid at levels sufficient to achieve a solution pH of 2.8 to 3.2.

(c) The additive is used as an antimicrobial agent in accordance with current industry practice in the processing of red meat, red meat parts, and organs as a component of a spray or in the processing of red meat parts and organs as a component of a dip. Applied as a dip or spray, the additive is used at levels that result in sodium chlorite concentrations between 500 and 1,200 ppm in combination with any GRAS acid at levels sufficient to achieve a solution pH of 2.5 to 2.9.

(d) The additive is used as an antimicrobial agent in water and ice that are used to rinse, wash, thaw, transport, or store seafood in accordance with current industry standards of good manufacturing practice. The additive is produced by mixing an aqueous solution of sodium chlorite with any GRAS acid to achieve a pH in the range of 2.5 to 2.9 and diluting this solution with water to achieve an actual use concentration of 40 to 50 parts per million (ppm) sodium chlorite. Any seafood that is intended to be consumed raw shall be subjected to a potable water rinse prior to consumption.

(e) The additive is used as an antimicrobial agent on raw agricultural commodities in the preparing, packing, or holding of the food for commercial purposes, consistent with section 201(q)(1)(B)(i) of the act, and not applied for use under section 201(q)(1)(B)(i)(II), (q)(1)(B)(i)(II), or (q)(1)(B)(i)(III) of the act, in accordance with current industry standards of good manufacturing practice. Applied as a dip or a spray, the additive is used at levels that result in chlorite concentrations of 500 to 1200 parts per million (ppm), in combination with any GRAS acid at levels sufficient to achieve a pH of 2.3 to 2.9. Treatment of the raw agricultural commodities with acidified sodium chlorite solutions shall be followed by a potable water rinse, or by blanching, cooking, or canning.

(f) The concentration of sodium chlorite is determined by a method entitled “Determination of Sodium Chlorite: 50 ppm to 1500 ppm Concentration,” September 13, 1995, developed by Alcide VerDate 27<APR>2000 14:28 May 10, 2000 Jkt 190063 PO 00000 Frm 00133 Fmt 8010 Sfmt 8010 Y:\SGML\190063T.XXX pfrm02 PsN: 190063T
§ 173.340 Defoaming agents.

Defoaming agents may be safely used in processing foods, in accordance with the following conditions:

(a) They consist of one or more of the following:

1. Substances generally recognized by qualified experts as safe in food or covered by prior sanctions for the use prescribed by this section.

2. Substances listed in this paragraph (a)(2) of this section, subject to any limitations imposed:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Dimethylpolysiloxane (substantially free from hydrolyzable chloride and alkoxy groups; no more than 18 percent loss in weight after heating 4 hours at 200 °C; viscosity 300 to 1,050 centistokes at 25 °C; refractive index 1.400–1.404 at 25 °C).</td>
<td>10 parts per million in food, or at such level in a concentrated food that when prepared as directed on the labels, the food in its ready-for-consumption state will have not more than 10 parts per million except as follows: Zero in milk; 110 parts per million in dry gelatin dessert mixes labeled for use whereby no more than 16 parts per million is present in the ready-to-serve dessert; 250 parts per million in salt labeled for cooking purposes, whereby no more than 10 parts per million is present in the cooked food. As a preservative in defoaming agents containing dimethylpolysiloxane, in an amount not exceeding 1.0 percent of the dimethylpolysiloxane content. For use as prescribed in §172.808(h)(3) of this chapter.</td>
</tr>
<tr>
<td>Formaldehyde</td>
<td>As a stabilizer and thickener in defoaming agents containing dimethylpolysiloxane in an amount reasonably required to accomplish the intended effect. As defined in §172.820 of this chapter.</td>
</tr>
<tr>
<td>α-Hydro-omega-hydroxy-poly (oxyethylene)poly(oxypropylene) (minimum 15 moles)/poly(oxyethylene) block copolymer (CAS Reg. No. 9003–11–6) as defined in §172.808(a)(3) of this chapter.</td>
<td>As defined in §172.808 of this chapter.</td>
</tr>
<tr>
<td>Polyacrylic acid, sodium salt</td>
<td>As defined in §172.838 of this chapter.</td>
</tr>
<tr>
<td>Polyethylene glycol</td>
<td>As defined in U.S.P. XVI.</td>
</tr>
<tr>
<td>Polysorbate 60</td>
<td>As defined in §172.836 of this chapter.</td>
</tr>
<tr>
<td>Polysorbate 65</td>
<td>As defined in §172.838 of this chapter.</td>
</tr>
<tr>
<td>Propylene glycol alginate</td>
<td>As defined in §172.858 of this chapter.</td>
</tr>
<tr>
<td>Silicon dioxide</td>
<td>As defined in §172.840 of this chapter.</td>
</tr>
<tr>
<td>Sorbitan monostearate</td>
<td>As defined in §172.842 of this chapter.</td>
</tr>
<tr>
<td>White mineral oil: Conforming with §172.878 of this chapter...</td>
<td>As a component of defoaming agents for use in wash water for sliced potatoes at a level not to exceed 0.008 percent of the wash water.</td>
</tr>
</tbody>
</table>

(3) Substances listed in this paragraph (a)(3), provided they are components of defoaming agents limited to use in processing beet sugar and yeast, and subject to any limitations imposed:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aluminum stearate</td>
<td>As defined in §172.863 of this chapter.</td>
</tr>
<tr>
<td>Butyl stearate</td>
<td>As defined in §172.863 of this chapter.</td>
</tr>
<tr>
<td>BHA</td>
<td>As an antioxidant, not to exceed 0.1 percent by weight of defoamer.</td>
</tr>
<tr>
<td>BHT</td>
<td>Do.</td>
</tr>
<tr>
<td>Calcium stearate</td>
<td>As defined in §172.863 of this chapter.</td>
</tr>
<tr>
<td>Fatty acids</td>
<td>As defined in §172.860 of this chapter.</td>
</tr>
<tr>
<td>Formaldehyde</td>
<td>As a preservative.</td>
</tr>
<tr>
<td>Hydroxylated lecithin</td>
<td>As defined in §172.814 of this chapter.</td>
</tr>
<tr>
<td>Isopropl alcohol</td>
<td>As defined in §172.863 of this chapter.</td>
</tr>
<tr>
<td>Magnesium stearate</td>
<td>As defined in §172.863 of this chapter.</td>
</tr>
</tbody>
</table>
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<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Mineral oil: Conforming with § 172.878 of this chapter</td>
<td>Not more than 150 p.p.m. in yeast, measured as hydrocarbons.</td>
</tr>
<tr>
<td>Odorless light petroleum hydrocarbons: Conforming with § 172.884 of this chapter.</td>
<td>As an emulsifier not to exceed 10 percent by weight of defoamer formulation.</td>
</tr>
<tr>
<td>Petroleum wax: Conforming with § 172.880 of this chapter.</td>
<td>Complying with § 172.862 of this chapter.</td>
</tr>
<tr>
<td>Petroleum wax, synthetic.</td>
<td>As defined in § 172.863 of this chapter.</td>
</tr>
<tr>
<td>Polyethylene glycol (400)dioleate: Conforming with § 172.820(a)(2) of this chapter and providing the oleic acid used in the production of this substance complies with § 172.860 or § 172.862 of this chapter.</td>
<td>As defined in § 172.840 of this chapter.</td>
</tr>
<tr>
<td>Oleic acid derived from tall oil fatty acids</td>
<td>As defined in § 172.862 of this chapter.</td>
</tr>
<tr>
<td>Oxystearin</td>
<td>As defined in § 172.818 of this chapter.</td>
</tr>
<tr>
<td>Poly oxyethylene (600) dioleate. Poly oxyethylene (600) monoricinoleate.</td>
<td>Molecular weight range, 1,200–3,000.</td>
</tr>
<tr>
<td>Polysorbate 80</td>
<td>As defined in § 172.840 of this chapter.</td>
</tr>
<tr>
<td>Potassium stearate</td>
<td>As defined in § 172.863 of this chapter.</td>
</tr>
<tr>
<td>Propylene glycol mono- and diesters of fats and fatty acids</td>
<td>As defined in § 172.856 of this chapter.</td>
</tr>
<tr>
<td>Soybean oil fatty acids, hydroxylated. Tallow, hydrogenated, oxidized or sulfated. Tallow alcohol, hydrogenated.</td>
<td>As defined in § 172.856 of this chapter.</td>
</tr>
</tbody>
</table>

(4) The substances listed in this paragraph (a)(4), provided they are components of defoaming agents limited to use in processing beet sugar only, and subject to the limitations imposed:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>n-Butoxypoly(oxyethylene)-poly(oxypropylene)glycol.</td>
<td>Viscosity range, 4,850–5,350 Saybolt Universal Seconds (SUS) at 37.8 °C (100 °F). The viscosity range is determined by the method &quot;Viscosity Determination of n-butoxypoly(oxyethylene)-poly(oxypropylene)glycol&quot; dated April 25, 1995, developed by Union Carbide Corp., P.O. Box 670, Bound Brook, NJ 08805, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies of the material incorporated by reference are available from the Division of Petition Control, Center for Food Safety and Applied Nutrition (HFS-215), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, and may be examined at the Center for Food Safety and Applied Nutrition's Library, 200 C St. SW., rm. 3221, Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC.</td>
</tr>
</tbody>
</table>

(b) They are added in an amount not in excess of that reasonably required to inhibit foaming.


#### § 173.342 Chlorofluorocarbon 113 and perfluorohexane.

A mixture of 99 percent chlorofluorocarbon 113 (1,1,2-trichloro-1,2,2-trifluoroethane) (CAS Reg. No. 76-13-1, also known as fluorocarbon 113, CFC 113 and FC 113) and 1 percent perfluorohexane (CAS Reg. No. 355-42-0) may be safely used in accordance with the following prescribed conditions:

(a) The additive chlorofluorocarbon 113 has a purity of not less than 99.99 percent.

(b) The additive mixture is intended for use to quickly cool or crust-freeze chickens sealed in intact bags composed of substances regulated in parts 174, 175, 177, 178, and § 179.45 of this
§ 173.345 Chloropentafluoroethane.

The food additive chloropentafluoroethane may be safely used in food in accordance with the following prescribed conditions:

(a) The food additive has a purity of not less than 99.97 percent, and contains not more than 200 parts per million saturated fluoro compounds and 10 parts per million unsaturated fluoro compounds as impurities.

(b) The additive is used or intended for use alone or with one or more of the following substances: Carbon dioxide, nitrous oxide, propane, and octafluorocyclobutane complying with § 173.360, as an aerating agent for foamed or sprayed food products, with any propellant effect being incidental and no more than is minimally necessary to achieve the aerating function, except that use is not permitted for those standardized foods that do not provide for such use.

(c) To assure safe use of the additive in addition to the other information required by the act, the label or labeling of the food additive container shall bear adequate directions for use to provide a combustion product gas that complies with the limitations prescribed in paragraph (b) of this section, including instructions to assure proper filtration.

(d) The food additive is tested for compliance with paragraph (b)(2) by the following empirical method:

Spectrophotometric measurements. All measurements are made in an ultraviolet spectrophotometer in optical cells of 5 centimeters in length, and in the range of 255 millimicrons to 310 millimicrons, under the identical conditions of the test. The standard reference absorbance is the absorbance at 275 millimicrons of a standard reference solution of naphthalene (National Bureau of Standards Material No. 577 or equivalent in purity) containing a concentration of 1.4 milligrams per liter in purified isooctane, measured against isooctane of the same spectral purity in 5-centimeter cells. (This absorbance will be approximately 0.30.)

Solvent. The solvent used is pure grade isooctane having an ultraviolet absorbance not to exceed 0.02 measured against distilled water as a reference. Upon passage of purified inert gas through some isooctane under the identical conditions of the test, a lowering of the absorbance value has been observed. The absorbance of isooctane to be used in this procedure shall not be more than 0.02 lower in the range 255 millimicrons to 310 millimicrons, inclusive, than that of the

§ 173.350 Combustion product gas.

The food additive combustion product gas may be safely used in the processing and packaging of the foods designated in paragraph (c) of this section for the purpose of removing and displacing oxygen in accordance with the following prescribed conditions:

(a) The food additive is manufactured by the controlled combustion in air of butane, propane, or natural gas. The combustion equipment shall be provided with an absorption-type filter capable of removing possible toxic impurities, through which all gas used in the treatment of food shall pass; and with suitable controls to insure that any combustion products failing to meet the specifications provided in this section will be prevented from reaching the food being treated.

(b) The food additive meets the following specifications:

(1) Carbon monoxide content not to exceed 4.5 percent by volume.

(2) The ultraviolet absorbance in isooctane solution in the range 255 millimicrons to 310 millimicrons not to exceed one-third of the standard reference absorbance when tested as described in paragraph (e) of this section.

(c) It is used or intended for use to displace or remove oxygen in the processing, storage, or packaging of beverage products and other food, except fresh meats.

(d) To assure safe use of the additive in addition to the other information required by the act, the label or labeling of the combustion device shall bear adequate directions for use to provide a combustion product gas that complies with the limitations prescribed in paragraph (b) of this section, including instructions to assure proper filtration.

(e) The food additive is tested for compliance with paragraph (b)(2) by the following empirical method:

Spectrophotometric measurements. All measurements are made in an ultraviolet spectrophotometer in optical cells of 5 centimeters in length, and in the range of 255 millimicrons to 310 millimicrons, under the identical conditions of the test. The standard reference absorbance is the absorbance at 275 millimicrons of a standard reference solution of naphthalene (National Bureau of Standards Material No. 577 or equivalent in purity) containing a concentration of 1.4 milligrams per liter in purified isooctane, measured against isooctane of the same spectral purity in 5-centimeter cells. (This absorbance will be approximately 0.30.)
untreated solvent as measured in a 5-centimeter cell. If necessary to obtain the prescribed purities, the isooctane may be passed through activated silica gel.

Apparatus. To assure reproducible results, the additive is passed into the isooctane solution through a gas-absorption train consisting of the following components and necessary connections:

1. A gas flow meter with a range up to 30 liters per hour provided with a constant differential relay or other device to maintain a constant flow rate independent of the input pressure.

2. An absorption apparatus consisting of an inlet gas dispersion tube inserted to the bottom of a covered cylindrical vessel with a suitable outlet on the vessel for effluent gas. The dimensions and arrangement of tube and vessel are such that the inlet tube introduces the gas at a point not above 5½ inches below the surface of the solvent through a sintered glass outlet. The dimensions of the vessel are such, and both inlet and vessel are so designed, that the gas can be bubbled through 60 milliliters of isooctane solvent at a rate up to 30 liters per hour without mechanical loss of solvent. The level corresponding to 60 milliliters should be marked on the vessel.

3. A cooling bath containing crushed ice and water to permit immersion of the absorption vessel at least to the solvent level mark.

Caution. The various parts of the absorption train must be connected by gas-tight tubing and joints composed of materials which will neither remove components from nor add components to the gas stream. The gas source is connected in series to the flow-rate device, the flow meter, and the absorption apparatus in that order. Ventilation should be provided for the effluent gases which may contain carbon monoxide.

Sampling procedure. Immerse the gas-absorption apparatus containing 60 milliliters of isooctane in the coolant bath so that the solvent is completely immersed. Cool for at least 15 minutes and then pass 120 liters of the test gas through the absorption train at a rate of 30 liters per hour or less. Maintain the coolant bath at 0 °C throughout. Remove the absorption vessel from the bath, disconnect, and warm to room temperature. Add isooctane to bring the contents of the absorption vessel to 60 milliliters, and mix. Determine the absorbance of the solution in the 5-centimeter cell in the range 255 millimicrons to 310 millimicrons, inclusive, compared to isooctane. The absorbance of the solution of combustion product gas shall not exceed that of the isooctane solvent at any wavelength in the specified range by more than one-third of the standard reference absorbance.

§ 173.355 Dichlorodifluoromethane.

The food additive dichlorodifluoromethane may be safely used in food in accordance with the following prescribed conditions:

(a) The additive has a purity of not less than 99.97 percent.

(b) It is used or intended for use, in accordance with good manufacturing practice, as a direct-contact freezing agent for foods.

(c) To assure safe use of the additive:

1. The label of its container shall bear, in addition to the other information required by the act, the following:

   (i) The name of the additive, dichlorodifluoromethane, with or without the parenthetical name “Food Freezant 12’.

   (ii) The designation “food grade”.

2. The label or labeling of the food additive container shall bear adequate directions for use.

§ 173.357 Materials used as fixing agents in the immobilization of enzyme preparations.

Fixing agents may be safely used in the immobilization of enzyme preparations in accordance with the following conditions:

(a) The materials consist of one or more of the following:

1. Substances generally recognized as safe in food.

2. Substances identified in this subparagraph and subject to such limitations as are provided:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
<tbody>
<tr>
<td>Cellulose triacetate</td>
<td>May be used as a fixing material in the immobilization of lactase for use in reducing the lactose content of milk.</td>
</tr>
<tr>
<td>Diethylaminoethyl-cellulose</td>
<td>May be used as a fixing material in the immobilization of glucose isomerase enzyme preparations for use in the manufacture of high fructose corn syrup, in accordance with §184.1372 of this chapter.</td>
</tr>
<tr>
<td>Glutaraldehyde</td>
<td>Do.</td>
</tr>
<tr>
<td>Periodic acid (CAS Reg. No. 10450-60-9).</td>
<td>Do.</td>
</tr>
</tbody>
</table>
§ 173.360 Octafluorocyclobutane.

The food additive octafluorocyclobutane may be safely used as a propellant and aerating agent in foamed or sprayed food products in accordance with the following conditions:

(a) The food additive meets the following specifications:

<table>
<thead>
<tr>
<th>Substances</th>
<th>Limitations</th>
</tr>
</thead>
</table>
| Polyethylenimine reaction product with 1,2-dichloroethane (CAS Reg. No. 68130-97-2) is the reaction product of homopolymerization of ethylenimine in aqueous hydrochloric acid at 100 °C and of cross-linking with 1,2-dichloroethane. The finished polymer has an average molecular weight of 50,000 to 70,000 as determined by gel permeation chromatography. The analytical method is entitled “Methodology for Molecular Weight Detection of Polyethyleneimine,” which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies may be obtained from the Division of Petition Control, Center for Food Safety and Applied Nutrition (HFS-200), 200 C St. SW., Washington, DC 20204, and may be examined at the Center for Food Safety and Applied Nutrition’s Library, 200 C St. SW., rm. 3321, Washington, DC, or at the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC. | May be used as a fixing material in the immobilization of glucoamylase enzyme preparations from Aspergillus niger for use in the manufacture of beer. May be used as a fixing material in the immobilization of:
1. Glucose isomerase enzyme preparations for use in the manufacture of high fructose corn syrup, in accordance with § 184.1372 of this chapter.
2. Glucoamylase enzyme preparations from Aspergillus niger for use in the manufacture of beer. Residual ethylenimine in the finished polyethylenimine polymer will be less than 1 part per million as determined by gas chromatography-mass spectrometry. The residual ethylenimine is determined by an analytical method entitled “Methodology for Ethyleneimine Detection in Polyethyleneimine,” which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Residual 1,2-dichloroethane in the finished polyethyleneimine polymer will be less than 1 part per million as determined by gas chromatography. The residual 1,2-dichloroethane is determined by an analytical method entitled “Methodology for Ethylenedichloride Detection in Polyethyleneimine,” which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies may be obtained from the Division of Petition Control, Center for Food Safety and Applied Nutrition (HFS-215), 200 C St. SW., Washington, DC 20204, or may be examined at the Center for Food Safety and Applied Nutrition’s Library, 200 C St. SW., rm. 3321, Washington, DC, or the Office of the Federal Register, 800 North Capitol St. NW., suite 700, Washington, DC. |

(b) The fixed enzyme preparation is washed to remove residues of the fixing materials.

§ 173.385 Sodium methyl sulfate.

Sodium methyl sulfate may be present in pectin in accordance with the following conditions:

(a) It is present as the result of methylation of pectin by sulfuric acid and methyl alcohol and subsequent treatment with sodium bicarbonate.
(b) It does not exceed 0.1 percent by weight of the pectin.

§ 173.395 Trifluoromethane sulfonic acid.

Trifluoromethane sulfonic acid has the empirical formula CF₃SO₃H (CAS Reg. No. 1493-13-6). The catalyst (Trifluoromethane sulfonic acid) may safely be used in the production of cocoa butter substitute from palm oil (1-palmitoyl-2-oleoyl-3-stearin) (see § 184.1259 of this chapter) in accordance with the following conditions:

(a) The catalyst meets the following specifications:

- Appearance, Clear liquid.
- Color, Colorless to amber.
- Neutralization equivalent, 147-151.
- Water, 1 percent maximum.

(ii) The percentage of the additive present in the case of a mixture.
(iii) The designation “food grade”.
(ii) The label or labeling of the food additive container shall bear adequate directions for use.

Water, 1 percent maximum.
Fluoride ion, 0.03 percent maximum.
Heavy metals (as Pb), 30 parts per million maximum.
Arsenic (as As), 3 parts per million maximum.

(b) It is used at levels not to exceed 0.2 percent of the reaction mixture to catalyze the directed esterification.

(c) The esterification reaction is quenched with steam and water and the catalyst is removed with the aqueous phase. Final traces of catalyst are removed by washing batches of the product three times with an aqueous solution of 0.5 percent sodium bicarbonate.

(d) No residual catalyst may remain in the product at a detection limit of 0.2 part per million fluoride as determined by the method described in “Official Methods of Analysis of the Association of Official Analytical Chemists,” sections 25.049-25.055, 13th Ed. (1980), which is incorporated by reference. Copies may be obtained from the American Oil Chemists’ Society, 481 North Frederick Ave., suite 500, Gaithersburg, MD 20877-2504, or may be examined at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC 20408.

§ 173.400 Dimethyldialkylammonium chloride.

Dimethyldialkylammonium chloride may be safely used in food in accordance with the following prescribed conditions:

(a) The food additive is produced by one of the following methods:

(1) Ammonolysis of natural tallow fatty acids to form amines that are subsequently reacted with methyl chloride to form the quaternary ammonium compound known as dimethyl(2-ethylhexyl) hydrogenated tallow ammonium chloride and consisting primarily of dimethyl(2-ethylhexyl)octadecylammonium chloride and dimethyl(2-ethylhexyl)hexadecylammonium chloride.

(b) The food additive described in paragraph (a)(1) of this section contains not more than a total of 2 percent by weight of free amine and amine hydrochloride. The food additive described in paragraph (a)(2) of this section contains not more than 3 percent by weight, each, of free amine and amine hydrochloride as determined by A.O.C.S. method Te 3a-64, “Acid Value and Free Amine Value of Fatty Quaternary Ammonium Chlorides,” 2d printing including additions and revisions 1990, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 1. Copies are available from the American Oil Chemists’ Society, P.O. Box 5037, Station A, Champaign, IL 61820, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC.

(c) The food additive is used as a decolorizing agent in the clarification of refinery sugar liquors under the following limitations:

(1) The food additive described in paragraph (a)(1) of this section is added only at the defecation/clarification stage of sugar liquor refining in an amount not to exceed 700 parts per million by weight of sugar solids.

(2) The food additive described in paragraph (a)(2) of this section is used under the following conditions:

(i) The additive is adsorbed onto a support column composed of suitable polymers that are regulated for contact with aqueous food. Excess non-adsorbed additive shall be rinsed away with potable water prior to passage of sugar liquor through the column.

(ii) The residue of the additive in the decolorized sugar liquor prior to crystallization shall not exceed 1 part per million of sugar as determined by a
method entitled “Colorimetric Determination of Residual Quaternary Ammonium Compounds (Arquad HTL8) in Sugar and Sugar Solutions,” June 13, 1990, which is incorporated by reference in accordance with 5 U.S.C. 552(a) and 1 CFR part 51. Copies are available from the Center for Food Safety and Applied Nutrition (HFS-200), Food and Drug Administration, 200 C St. SW., Washington, DC 20204, or available for inspection at the Office of the Federal Register, 800 North Capitol Street, NW., suite 700, Washington, DC.

(d) To assure safe use of the additive, the label and labeling of the additive shall bear, in addition to other information required by the Federal Food, Drug, and Cosmetic Act, adequate directions to assure use in compliance with paragraph (c) of this section.

[56 FR 42686, Aug. 29, 1991]

§ 174.6 Threshold of regulation for substances used in food-contact articles.

Substances used in food-contact articles (e.g., food-packaging or food-processing equipment) that migrate, or that may be expected to migrate, into food at negligible levels may be reviewed under §170.39 of this chapter. The Food and Drug Administration will exempt substances whose uses it determines meet the criteria in §170.39 of this chapter from regulation as food...