§ 91.312 Analytical gases.

(a) The shelf life of a calibration gas may not be exceeded. Record the expiration date stated by the gas supplier for each calibration gas.

(b) Engine test conditions. Measure the absolute temperature (designated as \( T \) and expressed in Kelvin) of the engine air at the inlet to the engine and the dry atmospheric pressure (designated as \( p \), and expressed in kPa. Determine the parameter \( f \) according to the following provisions:

1. Naturally aspirated and mechanically supercharged engines:

\[
f = \frac{99}{p_s} \left( \frac{T}{298} \right)^{0.7}
\]

2. Turbocharged engine with or without cooling of inlet air:

\[
f = \left( \frac{99}{p_s} \right)^{0.7} \left( \frac{T}{298} \right)^{1.5}
\]

(3) For a test to be recognized as valid, the parameter \( f \) must be between the limits as shown below:

\[0.96 < f < 1.04\]

§ 91.312 Analytical gases.

(a) The shelf life of a calibration gas may not be exceeded. Record the expiration date stated by the gas supplier for each calibration gas.

(b) Pure gases. The required purity of the gases is defined by the contamination limits given in parenthesis. The following gases must be available for operation.

1. Purified nitrogen, also referred to as “zero-grade nitrogen” (Contamination\(\leq 1\) ppm CO, \(\leq 400\) ppm CO\(_2\), \(\leq 0.1\) ppm NO)

2. Purified oxygen (Purity \(99.5\) percent vol \(O_2\))

3. Hydrogen-helium mixture (40±2 percent hydrogen, balance helium) (Contamination\(\leq 1\) ppm C, \(\leq 400\) ppm CO)

4. Purified synthetic air, also referred to as “zero gas” (Contamination\(\leq 1\) ppm CO, \(\leq 400\) ppm CO\(_2\), \(\leq 0.1\) ppm NO) (Oxygen content between 18–21 percent vol.)

(c) Calibration and span gases. (1) Calibration gas values are to be derived from NIST “Standard Reference Materials” (SRM’s) or other local gas standards and are to be single blends as specified in this subsection.

(2) Mixtures of gases having the following chemical compositions must be available:

- \(C_3H_8\) and purified synthetic air (dilute measurements);
- \(C_3H_8\) and purified nitrogen (raw measurements);
- CO and purified nitrogen;
- \(NO_2\) and purified nitrogen (the amount of NO\(_2\) contained in this calibration gas must not exceed five percent of the NO content);
- CO\(_2\) and purified nitrogen.

Note: For the HFID or FID, the manufacturer may choose to use a diluent span gas and the calibration gas either purified synthetic air or purified nitrogen. Any mixture of \(C_3H_8\) and purified synthetic air which contains a concentration of propane higher than what a gas supplier considers to be safe may be substituted with a mixture of \(C_2H_6\) and purified nitrogen. However, the manufacturer must be consistent in the choice of diluent (zero air or purified nitrogen) between the calibration and span gases. If a manufacturer chooses to use \(C_3H_8\) and purified nitrogen for the calibration gases, then purified nitrogen must be the diluent for the span gases.

(3) The true concentration of a span gas must be within ±2 percent of the NIST gas standard. The true concentration of a calibration gas must be within ±1 percent of the NIST gas standard. The use of precision blending devices (gas dividers) to obtain the required calibration gas concentrations is acceptable. Give all concentrations of calibration gas on a volume basis (volume percent or volume ppm).

(4) The gas concentrations used for calibration and span may also be obtained by means of a gas divider, diluting with purified \(N_2\) or with purified synthetic air. The accuracy of the mixing device must be such that the concentration of the diluted gases may be determined to within ±2 percent.

(d) Oxygen interference check gases must contain propane with 350 ppmC ±75 ppmC hydrocarbon. Determine the concentration value to calibration gas tolerances by chromatographic analysis of total hydrocarbons plus impurities or by dynamic blending. Use nitrogen as the predominant diluent with the balance oxygen.

(e) Fuel for the hydrocarbon flame ionization detector (HC-FID) must be a blend of 40±2 percent hydrogen with the balance being helium. The mixture...
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shall contain less than one ppm equivalent carbon response; 98 to 100 percent hydrogen fuel may be used with advance approval of the Administrator.

(f) Hydrocarbon analyzer burner air. The concentration of oxygen must be within one mole percent of the oxygen concentration of the burner air used in the latest oxygen interference check (percent O₂), see §91.316(d). If the difference in oxygen concentration is greater than one mole percent, then the oxygen interference must be checked and the analyzer adjusted if necessary, to meet the percent O₂ requirements. The burner air must contain less than two ppmC hydrocarbon.

§ 91.313 Analyzers required.

(a) Analyzers. Analyze measured gases with the following instruments:

(1) Carbon monoxide (CO) analysis. (i) The carbon monoxide analyzer must be of the non-dispersive infrared (NDIR) absorption type.

(ii) The use of linearizing circuits is permitted.

(2) Carbon dioxide (CO₂) analysis. (i) The carbon dioxide analyzer must be of the non-dispersive infrared (NDIR) absorption type.

(ii) The use of linearizing circuits is permitted.

(3) Oxygen (O₂) analysis. Oxygen (O₂) analyzers may be of the paramagnetic (PMD), zirconia (ZRDO) or electrochemical type (ECS).

(4) Hydrocarbon (HC) analysis. (i) For Raw Gas Sampling, the hydrocarbon analyzer must be of the heated flame ionization (HFID) type. For constant volume sampling, the hydrocarbon analyzer may be of the flame ionization (FID) type or of the heated flame ionization (HFID) type.

(ii) For the HFID system, if the temperature of the exhaust gas at the sample probe is below 190 °C, the temperature of the valves, pipe work, and so forth, must be controlled so as to maintain a wall temperature of 190 ±11 °C. If the temperature of the exhaust gas at the sample probe is above 190 °C, the temperature of the valves, pipe work, and so forth, must be controlled so as to maintain a wall temperature greater than 180 °C.

(iii) For the HFID analyzer, the detector, oven, and sample-handling components within the oven must be suitable for continuous operation at temperatures to 200 °C. It must be capable of maintaining temperature within ±5.5 °C of the set point.

(iv) Fuel and burner air must conform to the specifications in §91.312.

(v) The percent of oxygen interference must be less than three percent, as specified in §91.316(d).

(5) Oxides of nitrogen (NOX) analysis.

(i) This analysis device consists of the following items:

(A) A NO₂ to NO converter. The NO₂ to NO converter efficiency must be at least 90 percent.

(B) An ice bath located after the NOX converter (optional).

(C) A chemiluminescent detector (CLD) or heated chemiluminescent detector (HCLD).

(ii) The quench interference must be less than three percent as measured in §91.325.

(b) Other gas analyzers yielding equivalent results may be used with advance approval of the Administrator.

(c) The following requirements must be incorporated as indicated in systems used for testing under this subpart.

(1) Carbon monoxide and carbon dioxide measurements must be made on a dry basis (for raw exhaust measurement only). Specific requirements for the means of drying the sample can be found in §91.313(e).

(2) Calibration or span gases for the NOX measurement system must pass through the NO₂ to NO converter.

(d) The electromagnetic compatibility (EMC) of the equipment must be on a level as to minimize additional errors.

(e) Gas drying. Chemical dryers are not an acceptable method of removing water from the sample. Water removal by condensation is acceptable. If water is removed by condensation, the sample gas temperature or sample dew point must be monitored either within the water trap or downstream and its temperature must not exceed 7 °C. A water trap performing this function is an acceptable method. Means other than condensation may be used only with prior approval from the Administrator.